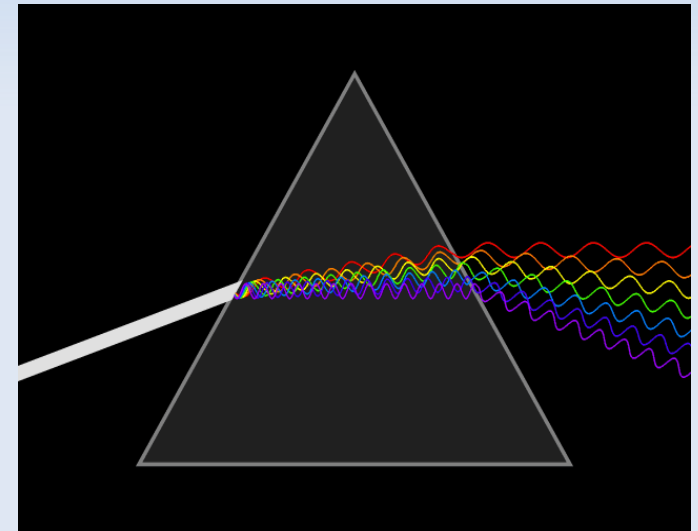


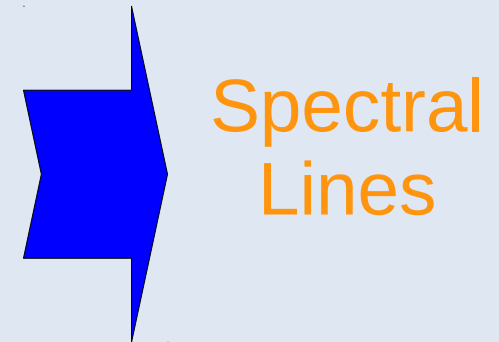
Spectral Line Observing

- Measurement goals
- Spectral line formation processes
- Line Shapes / Doppler effect
- Spectrometers
- Observing techniques
- Calibration
- Data reduction / Data products
- Data visualization



Measurement Goals

- What can we learn from radio spectral lines ?
- We can probe the physical, chemical and dynamical conditions of the interstellar matter (ISM) in the Milky Way and in external galaxies.
- Most ISM gas phases produce spectral lines:
 - Cold: 10 K, dense molecular gas (H_2)
 - Cool: 10^2 K, neutral gas (HI)
 - Warm: 10^4 K, ionized gas (HII)
 - Hot: 10^6 K, low-density ionized (SNR bubbles)



Measurement Goals (ctd.)

- Intensities can tell us about:
 - Gas temperature
 - Energy Sources
 - Gas density
 - Gravity / Cloud Criticality
 - Chemical composition
 - Abundances / Evolutionary State
 - Ionization / Magnetic Fields
 - Cloud Support



Measurement Goals (ctd.)

- Frequencies and line widths can be used to derive:
 - Dynamical models
 - Galaxy and Cloud Rotation
 - Cloud Collapse
 - Protostellar Outflows
 - Redshifts
 - Age
 - Distance



Spectroscopy

- Spectroscopy:

Any measurement of a quantity as a function of either wavelength (λ) or frequency (ν), i.e. also of energy ($E = h\nu$).

- Spectral Line:

Result of the interaction between a quantum system and a single photon.

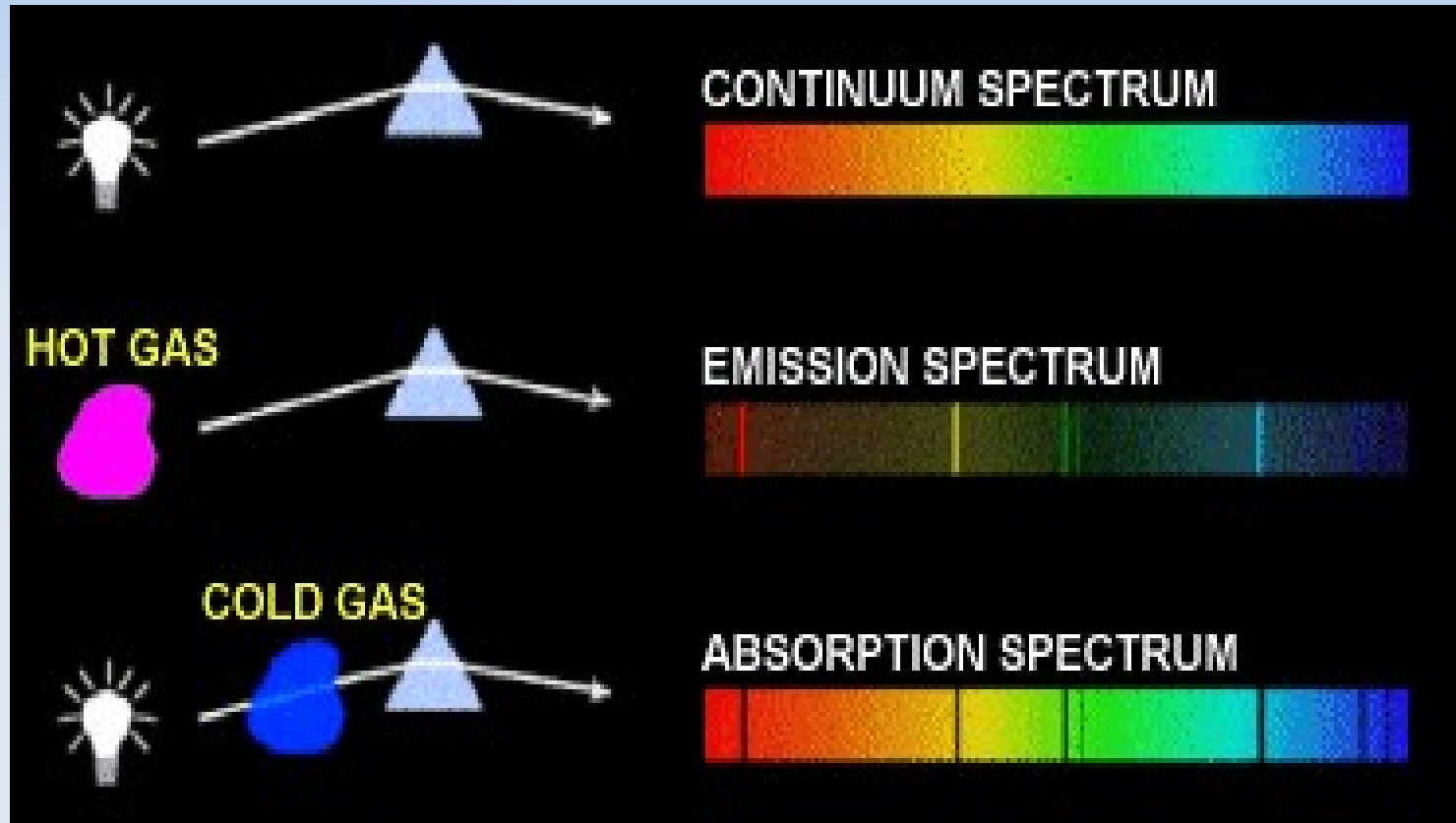


How do spectral lines form ?

- Quantum systems (atoms or molecules) can change their states only in discrete amounts of energy ΔE
- The transition between these states leads to emission or absorption of light at a single frequency $\nu = \Delta E/h$, the so called rest frequency
- Spectrally this transition is seen as a line



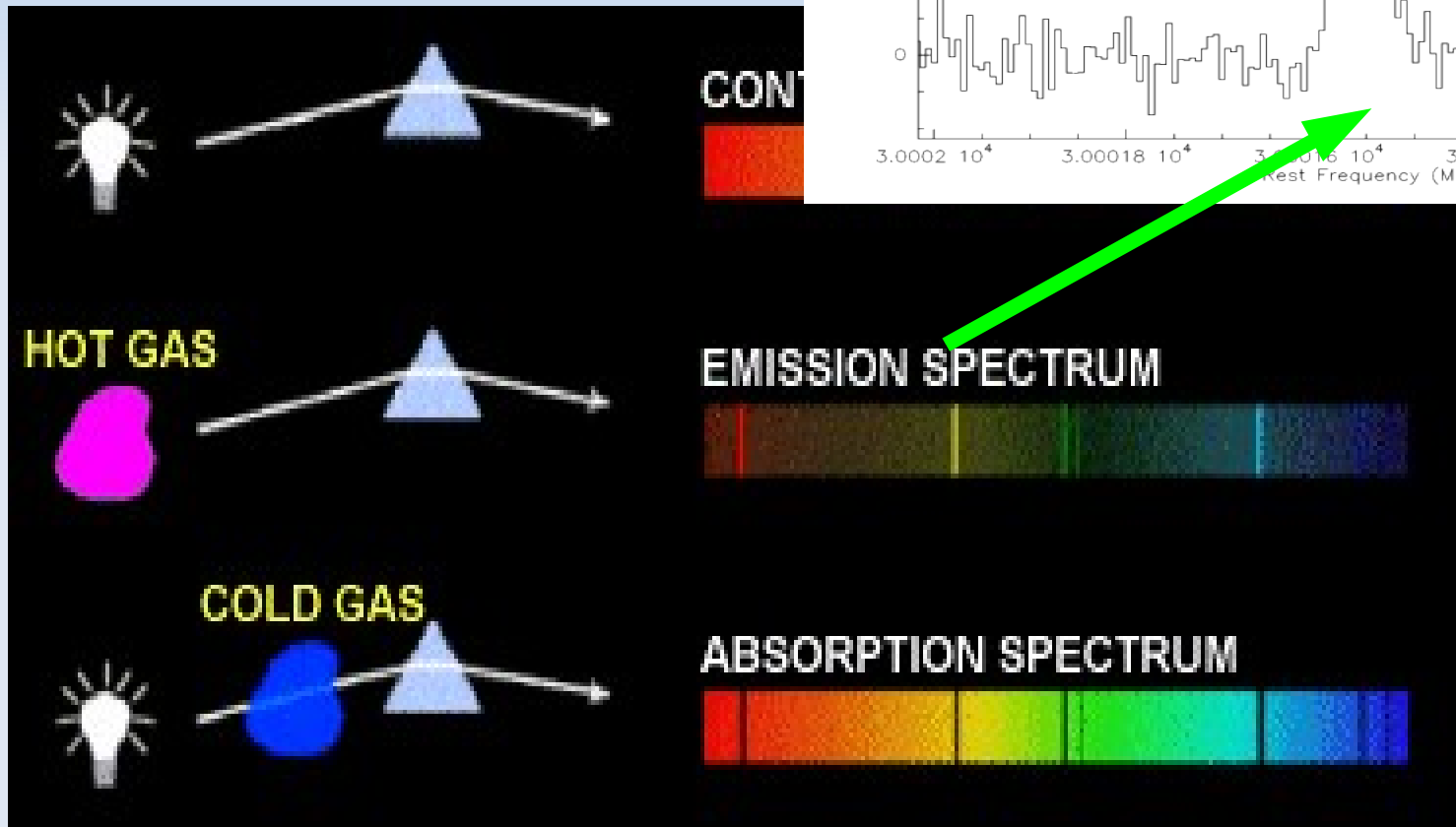
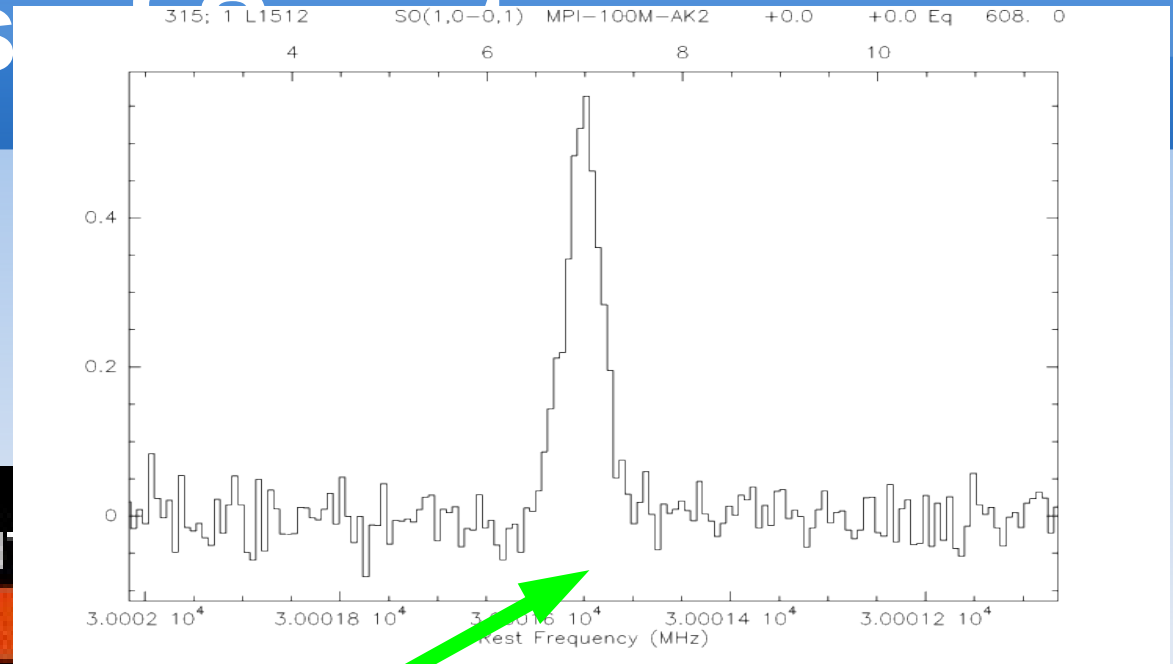
Types of Spectra



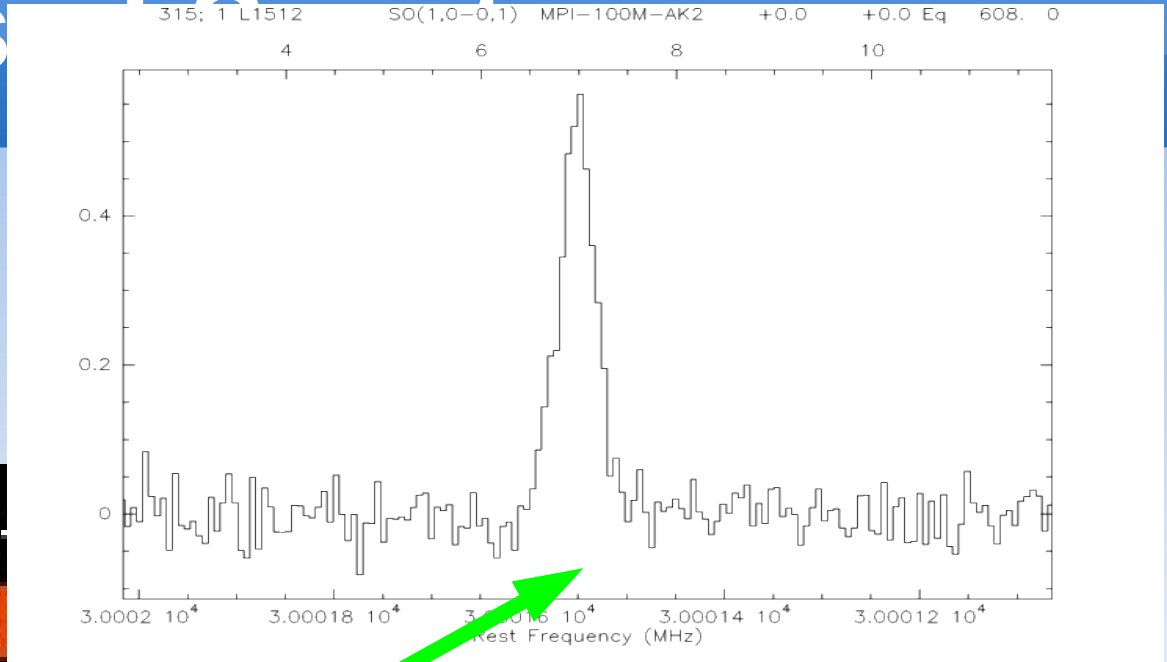
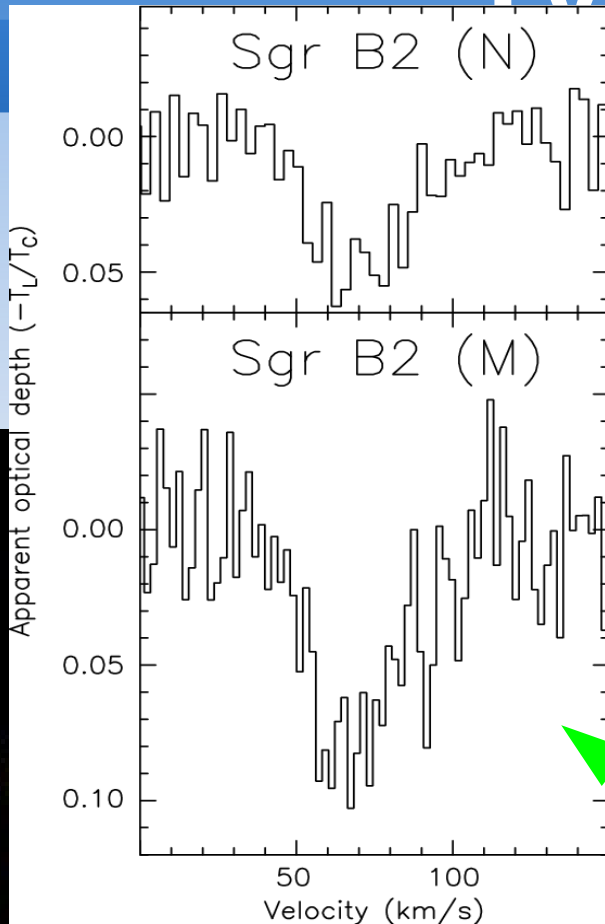
http://www.astro.columbia.edu/~archung/labs/fall2001/lec04_fall01.html



Types of Spectra



Types



CONT

EMISSION SPECTRUM

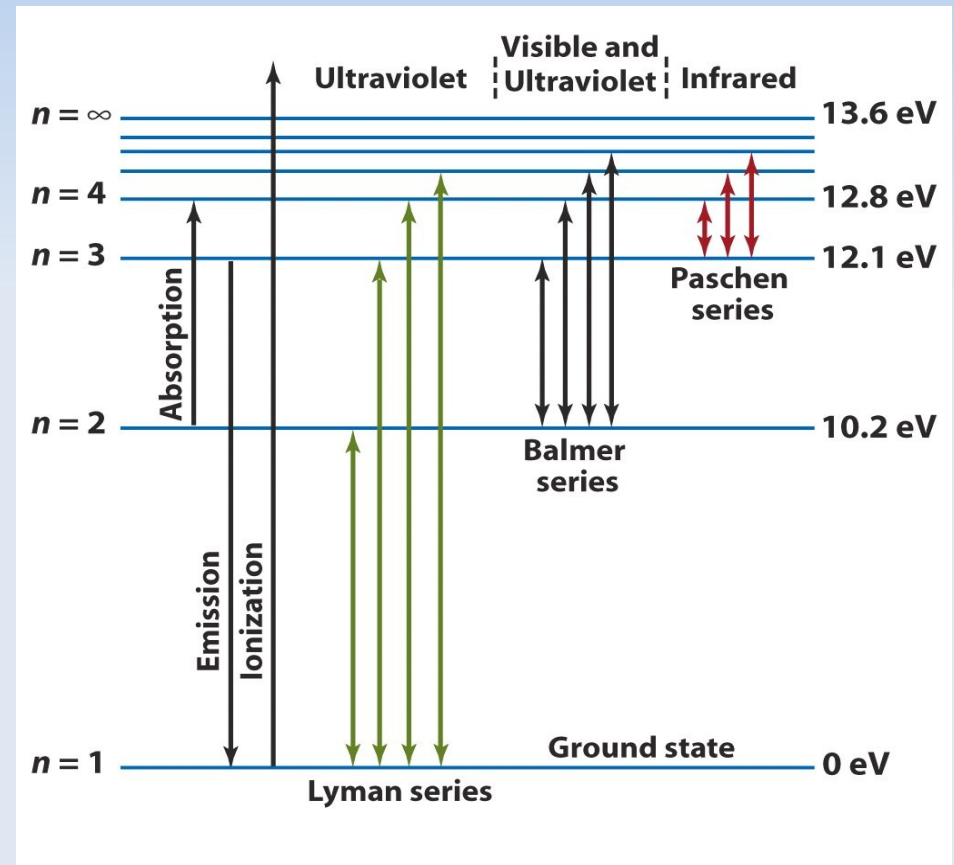
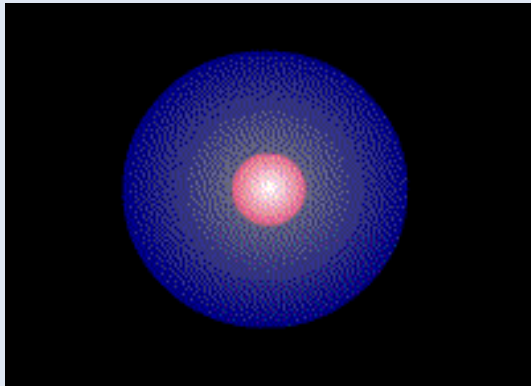
ABSORPTION SPECTRUM

COLD GAS



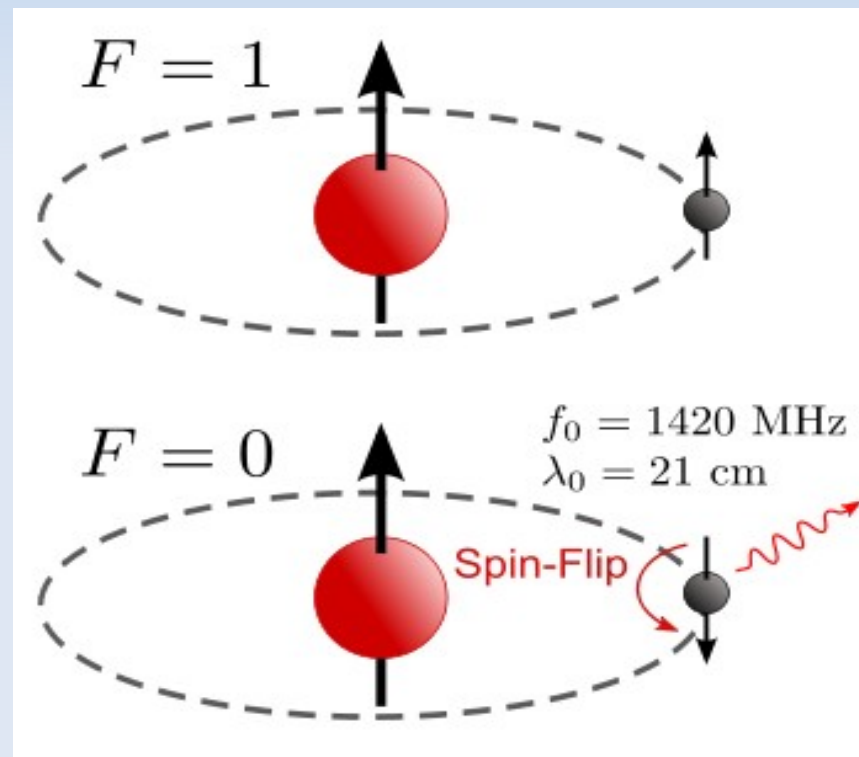
Atomic Lines I

- Electronic transitions (e.g. recombination lines ($H<n>\alpha$, etc.))



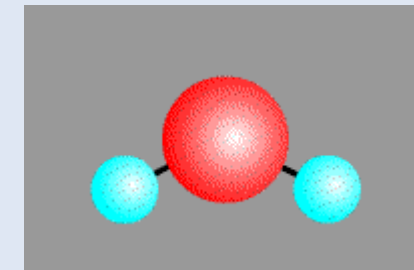
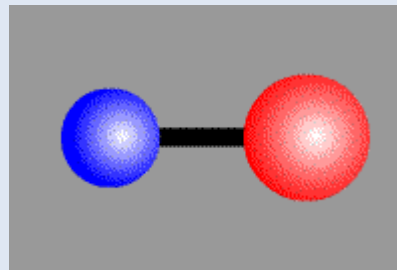
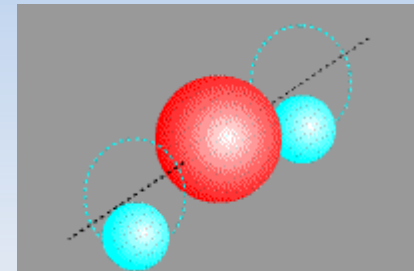
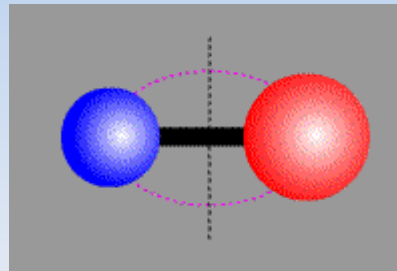
Atomic Lines II

- Hyperfine splitting / spin flips (e.g. HI 21 cm line
→ separate talk on Tuesday)



Molecular Lines I

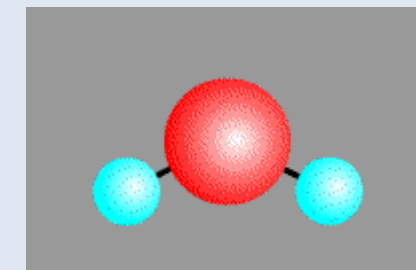
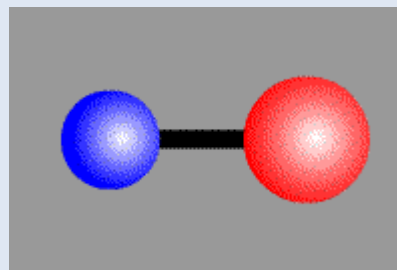
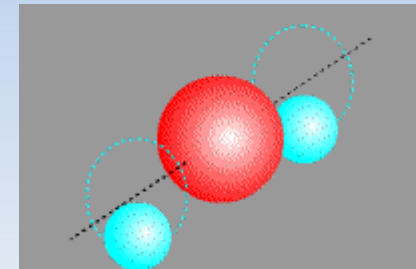
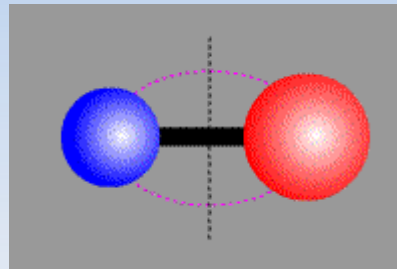
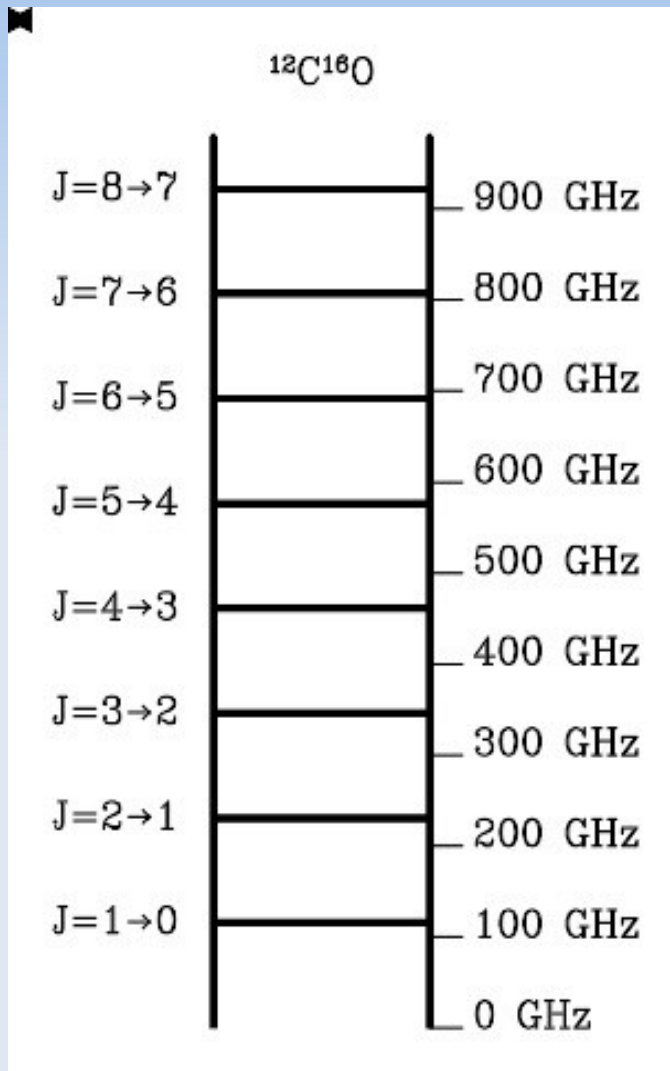
- Electronic transitions (rather in VIS / UV)
- Rotational transitions (needs dipole, so no H_2 !)
- Vibrational transitions



Animations: http://www.shokabo.co.jp/sp_e/optical/labo/opt_line/opt_line.htm



Molecular Lines I

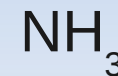
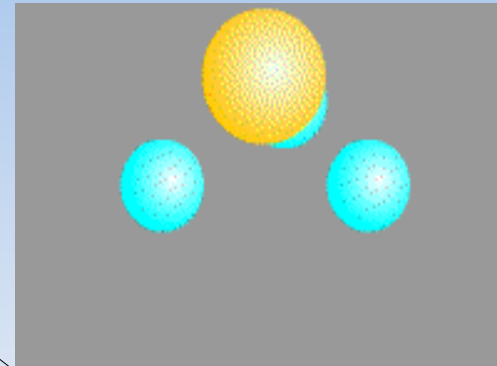


Animations: http://www.shokabo.co.jp/sp_e/optical/labo/opt_line/opt_line.htm

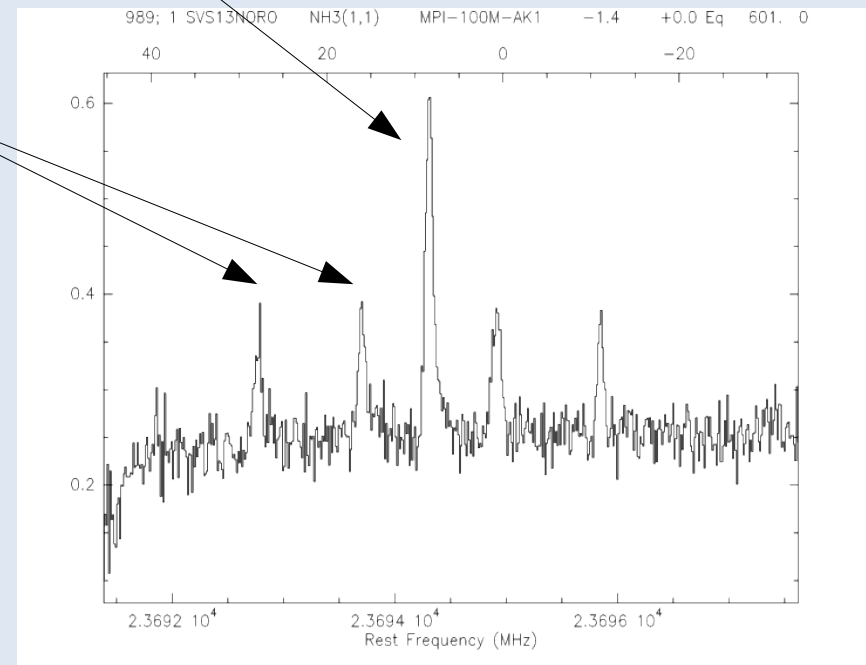
Molecular Lines II

- Inversion

Here: Nitrogen tunnels through double well potential

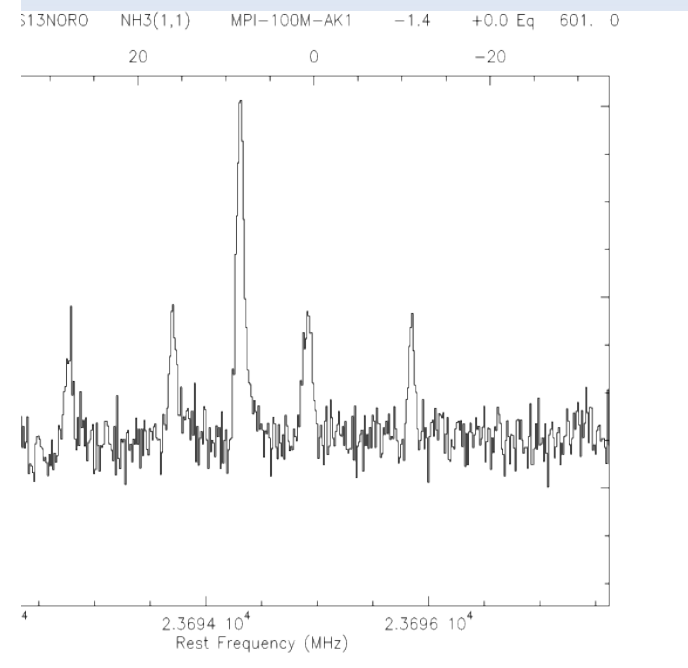
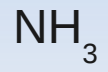
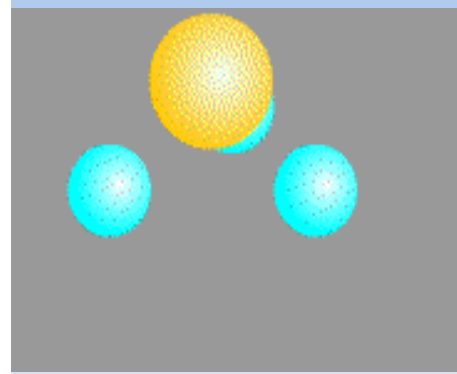
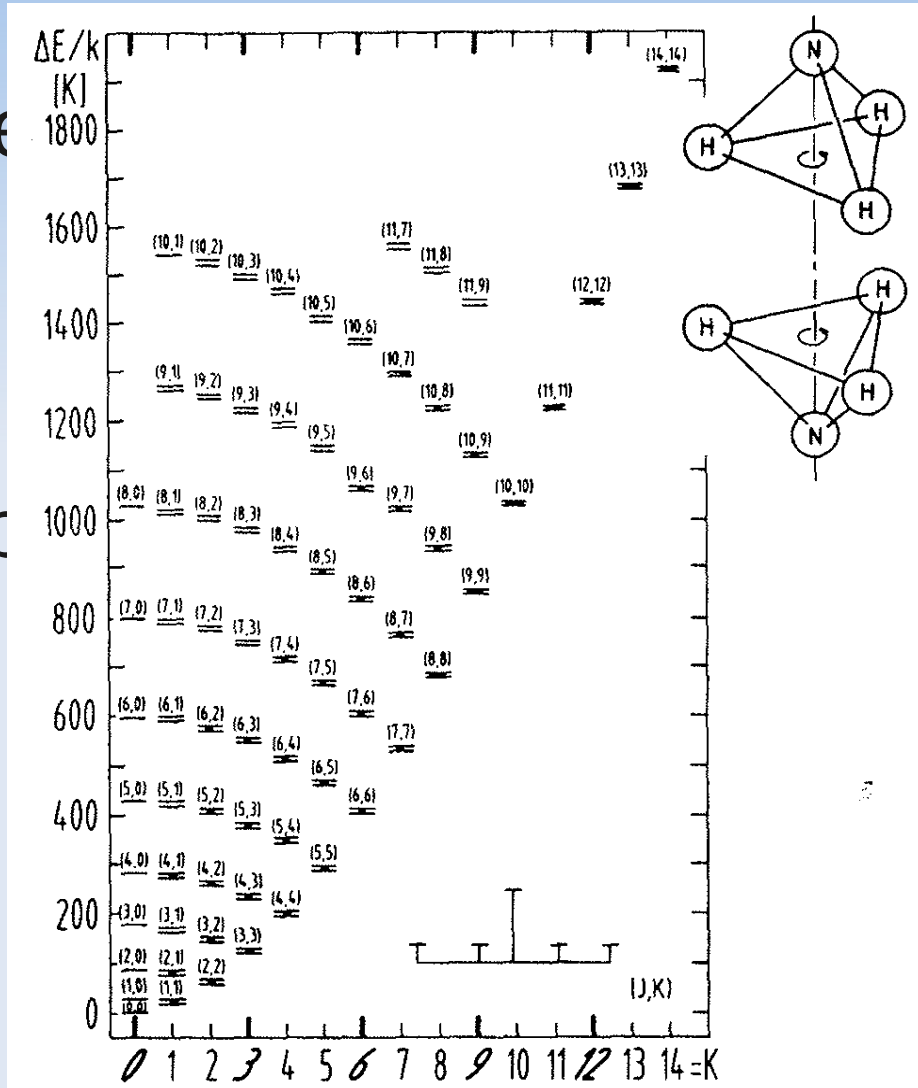


- Hyperfine splitting



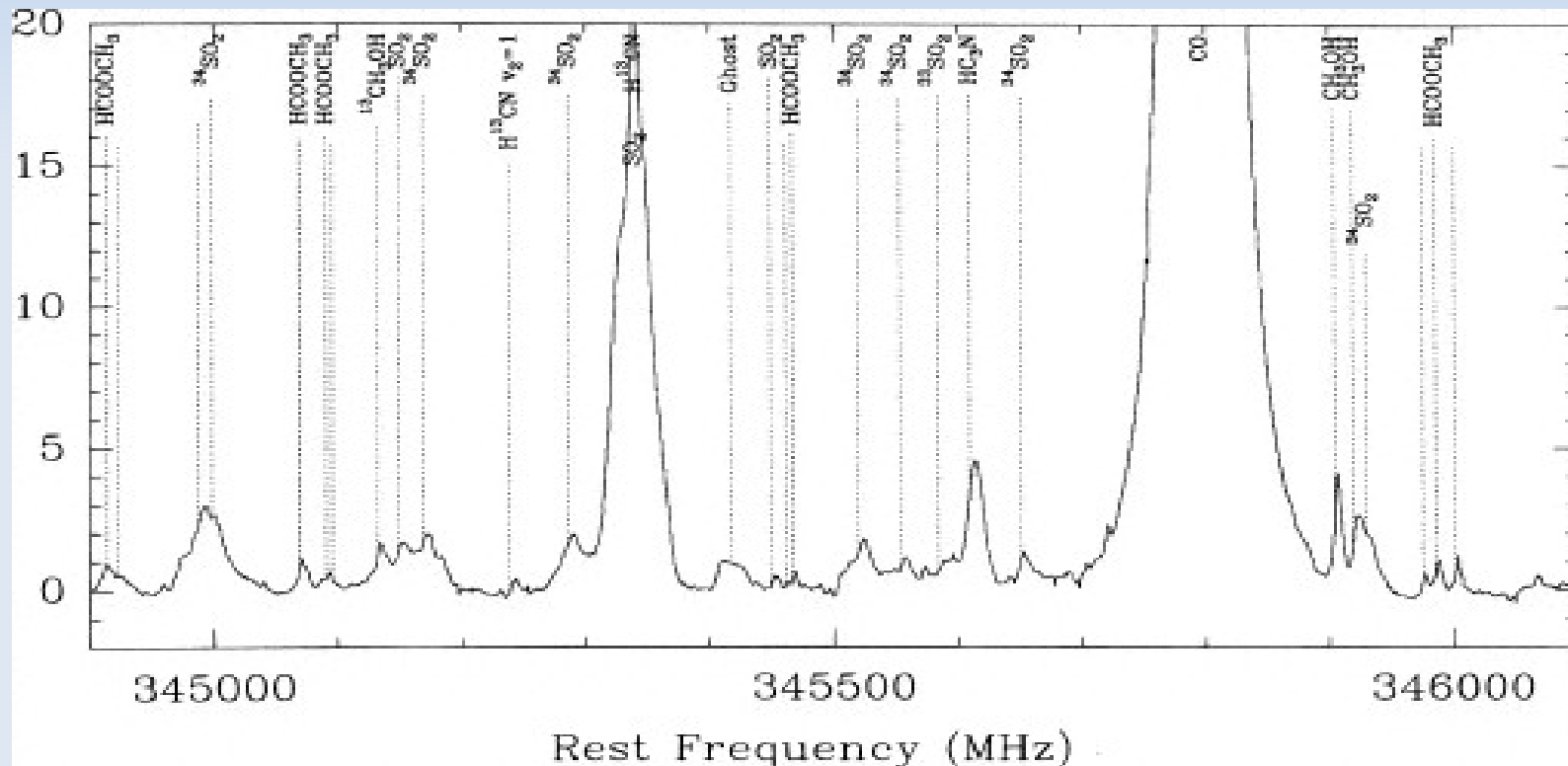
Molecular Lines II

- Inve
- Hyp



Interstellar Fingerprints

- Set of all possible lines of an atom or molecule is its personal “fingerprint”



Schilke et al. ApJS 108:301–337, 1997

Interstellar Molecular Zoo

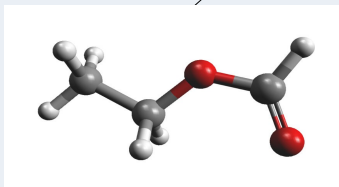
2 atoms	3 atoms	4 atoms	5 atoms	6 atoms	7 atoms	8 atoms	9 atoms	10 atoms	11 atoms	12 atoms	13 atoms
H ₂	C ₃ ⁺	c-C ₃ H	C ₅ ⁺	C ₆ H	C ₆ H 2008	CH ₃ C ₃ N	CH ₃ C ₄ H	CH ₃ C ₅ N	HC ₉ N	C ₆ H ₆ ⁺ (?)	HC ₁₁ N
AlF	C ₂ H	l-C ₃ H	C ₄ H	l-H ₂ C ₄	CH ₂ CHCN 2008	HC(O)OCH ₃	CH ₃ CH ₂ CN	(CH ₃) ₂ CO	CH ₃ C ₆ H	C ₂ H ₅ OCH ₃ ?	
AlCl	C ₂ O	C ₃ N	C ₄ Si	C ₂ H ₄ ⁺	CH ₃ C ₂ H	CH ₃ COOH	(CH ₃) ₂ O	(CH ₂ OH) ₂	C ₂ H ₅ OCHO 2009	n-C ₃ H ₇ CN 2009	
C ₂ ⁺⁺	C ₂ S	C ₃ O	l-C ₃ H ₂	CH ₃ CN	HC ₅ N	C ₇ H	CH ₃ CH ₂ OH	CH ₃ CH ₂ CHO			
CH	CH ₂	C ₃ S	c-C ₃ H ₂	CH ₃ NC	CH ₃ CHO	H ₂ C ₆	HC ₇ N				
CH ⁺	HCN	C ₂ H ₂ ⁺	H ₂ CCN	CH ₃ OH	CH ₃ NH ₂	CH ₂ OHCHO	C ₈ H				
CN	HCO	NH ₃	CH ₄ ⁺	CH ₃ SH	c-C ₂ H ₄ O	l-HC ₆ H ⁺ (?)	CH ₃ C(O)NH ₂				
CO	HCO ⁺	HCCN	HC ₃ N	HC ₃ NH ⁺	H ₂ CCHOH	CH ₂ CHCHO (?)	C ₈ H ⁻				
CO ⁺	HCS ⁺	HCNH ⁺	HC ₂ NC	HC ₂ CHO	C ₆ H ⁻	CH ₂ CCHCN	C ₃ H ₆				
CP	HOC ⁺	HNCO	HCOOH	NH ₂ CHO		H ₂ NCH ₂ CN 2008					
SiC	H ₂ O	HNCS	H ₂ CNH	C ₅ N							
HCl	H ₂ S	HOCO ⁺ 2008	H ₂ C ₂ O	l-HC ₄ H ⁺ (?)							
KCl	HNC	H ₂ CO	H ₂ NCN	l-HC ₄ N							
NH	HNO	H ₂ CN	HNC ₃	c-H ₂ C ₃ O							
NO	MgCN	H ₂ CS	SiH ₄ ⁺	H ₂ CCNH (?)							
NS	MgNC	H ₃ O ⁺	H ₂ COH ⁺	C ₅ N ⁻ 2008							
NaCl	N ₂ H ⁺	c-SiC ₃	C ₄ H ⁻ 2008	HC(O)CN 2008							
OH	N ₂ O	CH ₃ ⁺									
PN	NaCN	C ₃ N ⁻ 2008									
SO	OCS	PH ₃ ? 2008									
SO ⁺	SO ₂	HCNO 2009									
SiN	c-SiC ₂	HOCN 2010									
SiO	CO ₂ ⁺	HSCN 2009									
SiS	NH ₂										
CS	H ₃ ⁺⁺										
HF	H ₂ D ⁺ , HD ₂ ⁺										
HD	SiCN										
FeO ?	AlNC										
O ₂	SiNC										
CF ⁺	HCP										
SiH ?	CCP 2008										
PO	AlOH 2010										
AlO 2009	H ₂ O ⁺ 2010										
OH ⁺ 2010											
CN ⁻ 2010											

www.cdms.de 08/2010



Interstellar Molecular Zoo

2 atoms	3 atoms	4 atoms	5 atoms	6 atoms	7 atoms	8 atoms	9 atoms	10 atoms	11 atoms	12 atoms	13 atoms
H ₂	C ₃ ⁺	c-C ₃ H	C ₅ ⁺	C ₆ H	C ₆ H 2008	CH ₃ C ₃ N	CH ₃ C ₄ H	CH ₃ C ₅ N	HC ₉ N	C ₆ H ₆ ⁺ (?)	HC ₁₁ N
AlF	C ₂ H	l-C ₃ H	C ₄ H	l-H ₂ C ₄	CH ₂ CHCN 2008	HC(O)OCH ₃	CH ₃ CH ₂ CN	(CH ₃) ₂ CO	CH ₃ C ₆ H	C ₂ H ₅ OCH ₃ ?	
AlCl	C ₂ O	C ₃ N	C ₄ Si	C ₂ H ₄ ⁺	CH ₃ C ₂ H	CH ₃ COOH	(CH ₃) ₂ O	(CH ₂ OH) ₂	C ₂ H ₅ OCHO 2009	n-C ₃ H ₇ CN 2009	1
C ₂ ⁺⁺	C ₂ S	C ₃ O	l-C ₃ H ₂	CH ₃ CN	HC ₅ N	C ₇ H	CH ₃ CH ₂ OH	CH ₃ CH ₂ CHO			
CH	CH ₂	C ₃ S	c-C ₃ H ₂	CH ₃ NC	CH ₃ CHO	H ₂ C ₆	HC ₇ N				
CH ⁺	HCN	C ₂ H ₂ ⁺	H ₂ CCN	CH ₃ OH	CH ₃ NH ₂	CH ₂ OHCHO	C ₈ H		4	3	3
CN	HCO	NH ₃	CH ₄ ⁺	CH ₃ SH	c-C ₂ H ₄ O	l-HC ₆ H ⁺ (?)	CH ₃ C(O)NH ₂				
CO	HCO ⁺	HCCN	HC ₃ N	HC ₃ NH ⁺	H ₂ CCHOH	CH ₂ CHCHO (?)	C ₈ H ⁻				
CO ⁺	HCS ⁺	HCNH ⁺	HC ₂ NC	HC ₂ CHO	C ₆ H ⁻	CH ₂ CCHCN	C ₃ H ₆				
CP	HOC ⁺	HNCO	HCOOH	NH ₂ CHO		H ₂ NCH ₂ CN 2008					
SiC	H ₂ O	HNCS	H ₂ CNH	C ₅ N							
HCl	H ₂ S	HOCO ⁺ 2008	H ₂ C ₂ O	l-HC ₄ H ⁺ (?)	9	10	9				
KCl	HNC	H ₂ CO	H ₂ NCN	l-HC ₄ N							
NH	HNO	H ₂ CN	HNC ₃	c-H ₂ C ₃ O							
NO	MgCN	H ₂ CS	SiH ₄ ⁺	H ₂ CCNH (?)							
NS	MgNC	H ₃ O ⁺	H ₂ COH ⁺	C ₅ N ⁻ 2008							
NaCl	N ₂ H ⁺	c-SiC ₃	C ₄ H ⁻ 2008								
OH	N ₂ O	CH ₃ ⁺	HC(O)CN 2008	16							
PN	NaCN	C ₃ N ⁻ 2008		18							
SO	OCS	PH ₃ ? 2008									
SO ⁺	SO ₂	HCNO 2009									
SiN	c-SiC ₂	HOCN 2010									
SiO	CO ₂ ⁺	HSCN 2009									
SiS	NH ₂										
CS	H ₃ ⁺⁺	23									
HF	H ₂ D ⁺ , HD ₂ ⁺										
HD	SiCN										
FeO ?	AlNC										
O ₂	SiNC										
CF ⁺	HCP										
SiH ?	CCP 2008										
PO	AlOH 2010										
AlO 2009	H ₂ O ⁺ 2010										
OH ⁺ 2010											
CN ⁻ 2010											
35	33										



Ethyl formate

Belloche et al. A&A, 499, 215, 2009



Total of 164 and counting !

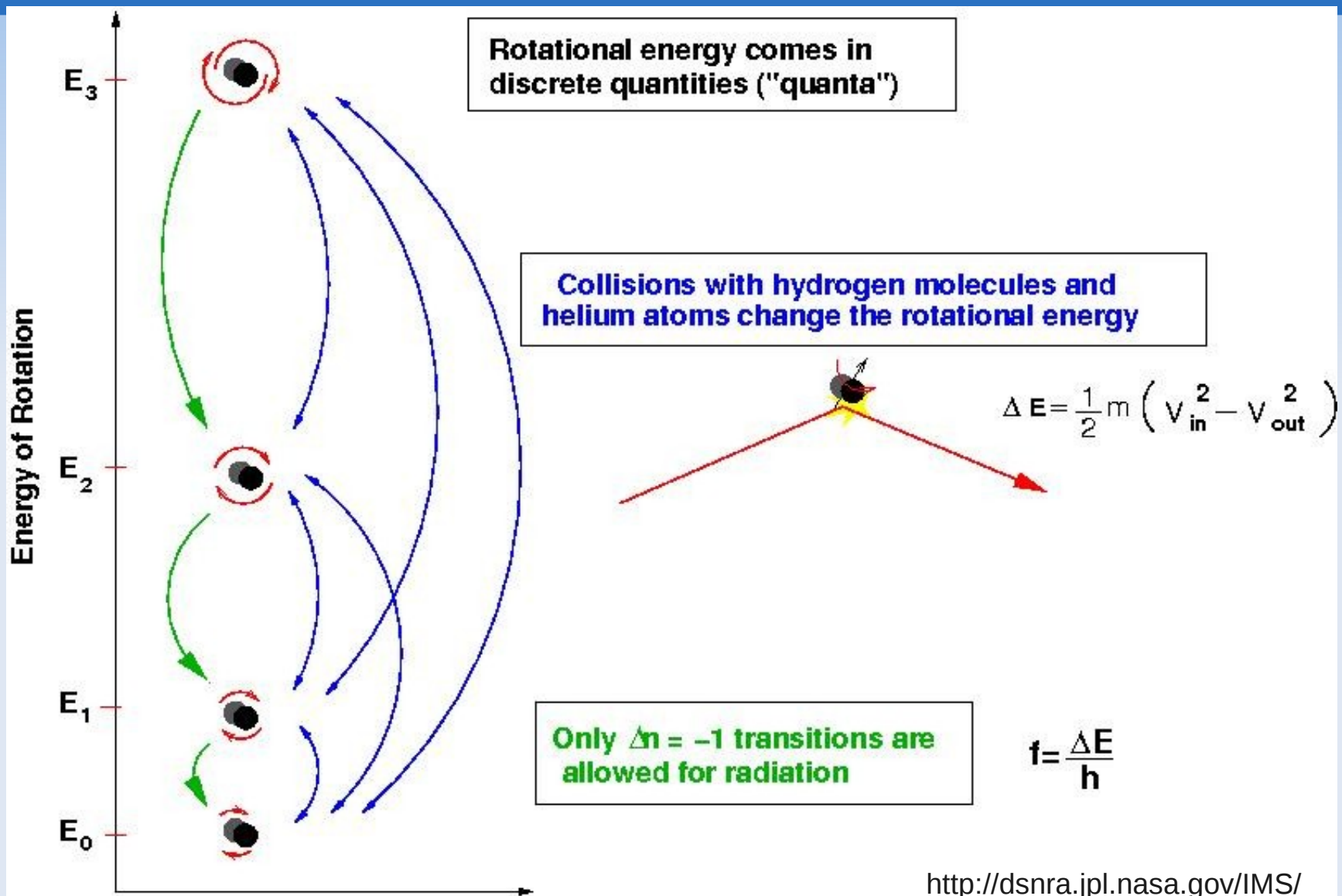
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Spectral Line Excitation

- Near HII regions radiatively via UV fields
- In cold molecular clouds via CMB but predominantly via collisions with H_2
- In case of a level inversion one gets a maser (→ see also special maser talk tomorrow)



Collisional Excitation



Maser Molecules

Molecules that exhibit maser action in celestial objects			
molecule	name	frequency (GHz)	characteristics*
OH	hydroxyl	1.612	O, M
	"	1.667	O, M
	"	1.720	O
H ₂ CO		4.829	O
CH ₃ OH	methanol	12.178	O
SiS	silicon sulfide	18.155	C
H ₂ O	water	22.235	O, M
NH ₃	ammonia	23.870	O
SiO	silicon oxide	43.122	M, S, O
	"	86.243	M, S
HCN	hydrogen cyanide	89.087	C

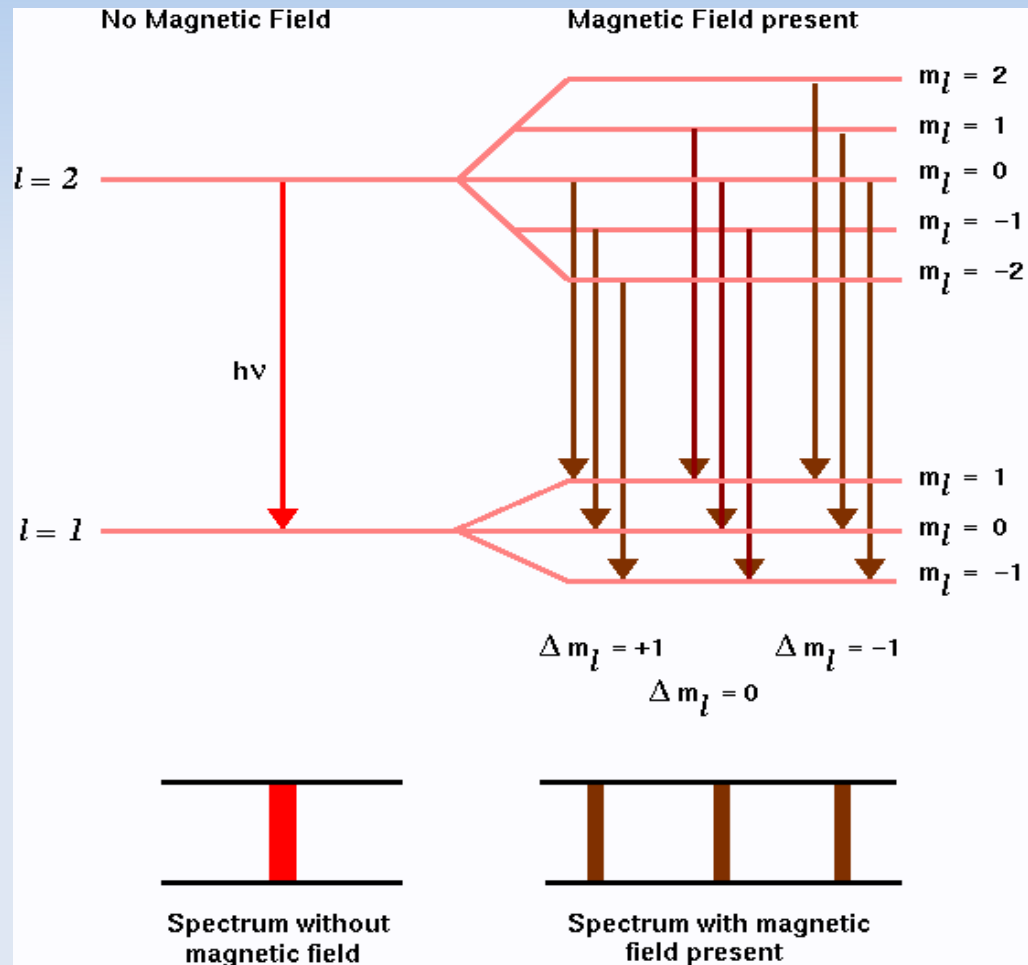
*O means that the maser emission is frequently found in star-forming regions; M, in M stars; S, in S stars; C, in carbon stars

http://www.daviddarling.info/encyclopedia/I/interstellar_maser.html



Zeeman Effect

- Degenerate energy levels split up if an external magnetic field is applied
- This leads to additional transitions and allows to measure the magnetic field



Kingshuk Majumdar (2000)



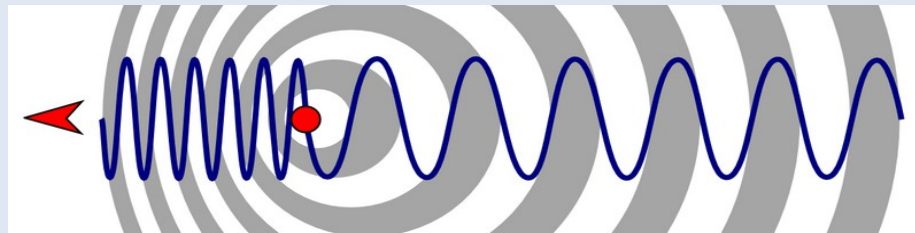
Optical Depth Effects

- Depending on the density and temperature spectral line emission can be optically thin or thick
- In the case of optical depth $\tau \ll 1$, one can look through a cloud and determine column densities and internal dynamics
- For $\tau \gg 1$, one can see only the surface of an object. Using radiative transfer one can calculate the cloud temperature



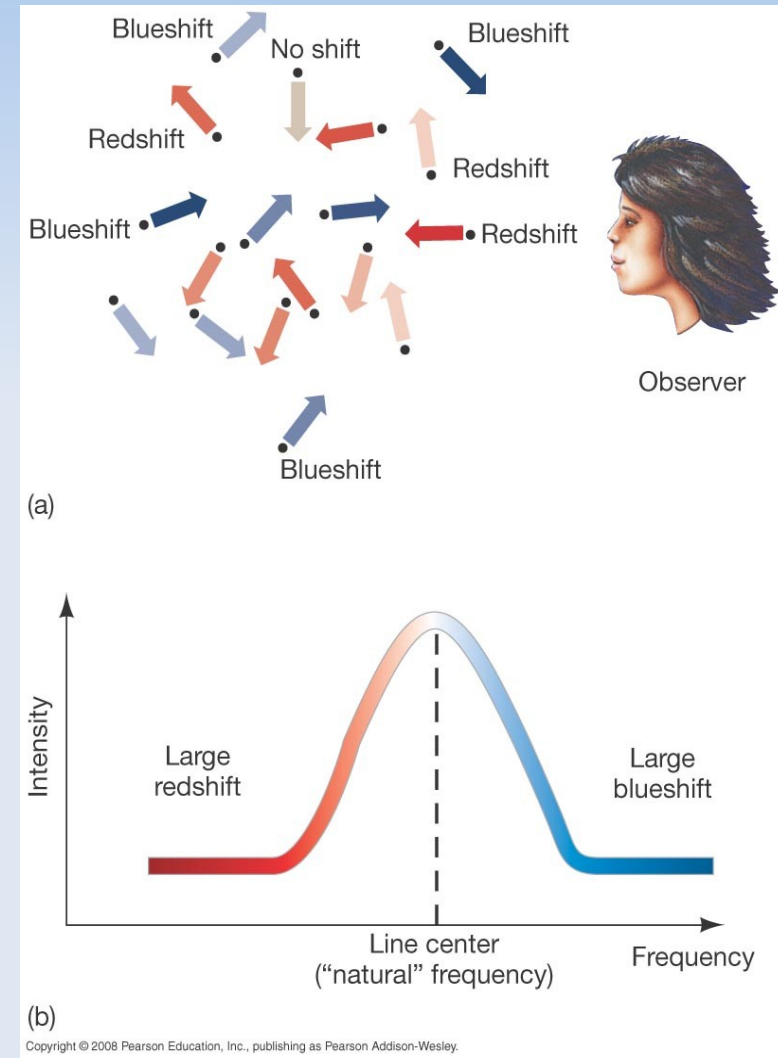
Line Profile Shape

- Ideal line should be infinitely sharp because there is a fixed energy difference $\Delta E = h\nu_0$
- Energy uncertainty causes a small broadening, the “natural line width”
- Thermal motion of emitters leads to Doppler shifted line frequencies



Thermal Broadening

- Considering statistical ensembles one can derive a Gaussian shape for the broadened line
- Only the line-of-sight, i.e. the radial component adds to this effect

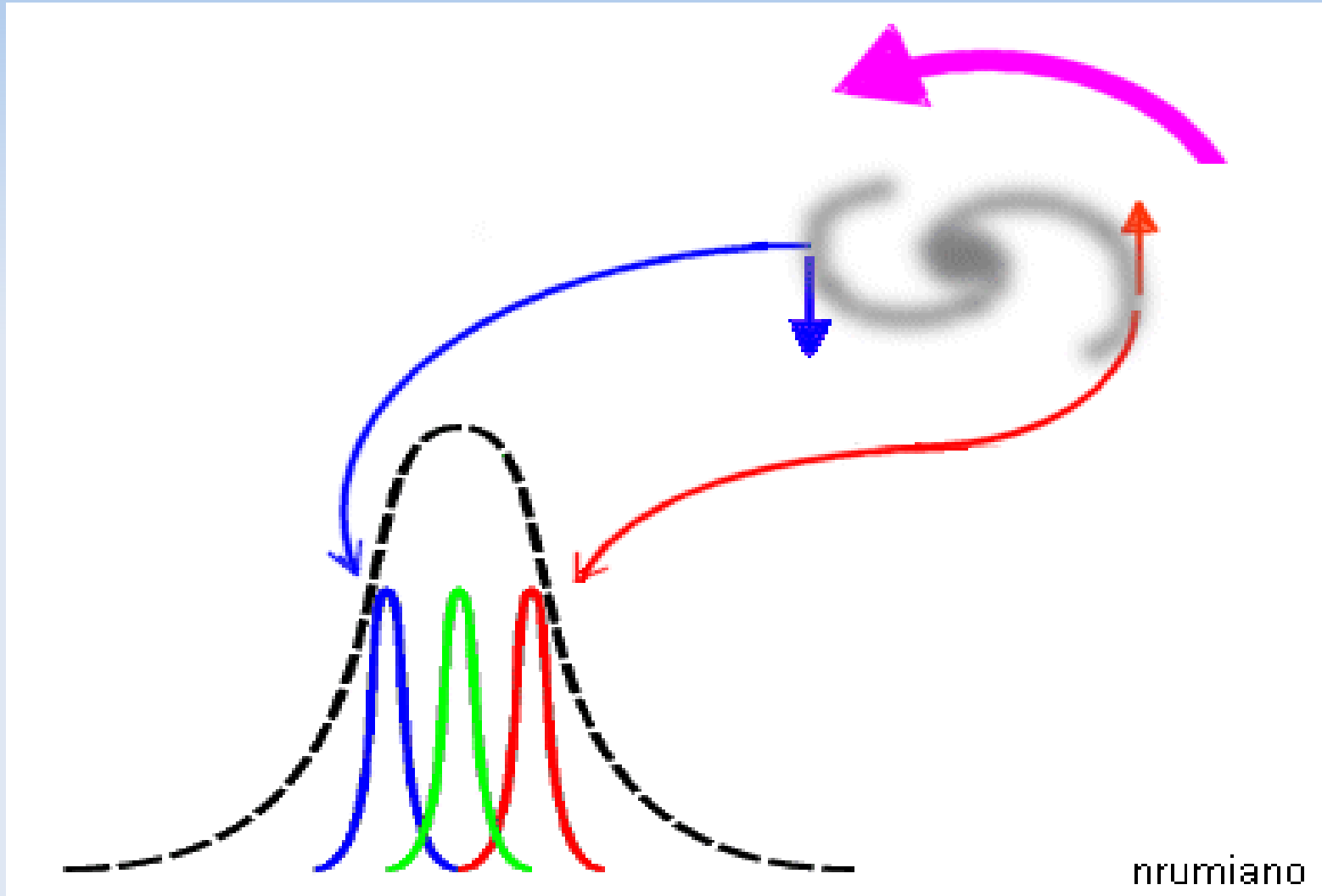


Radial Velocity

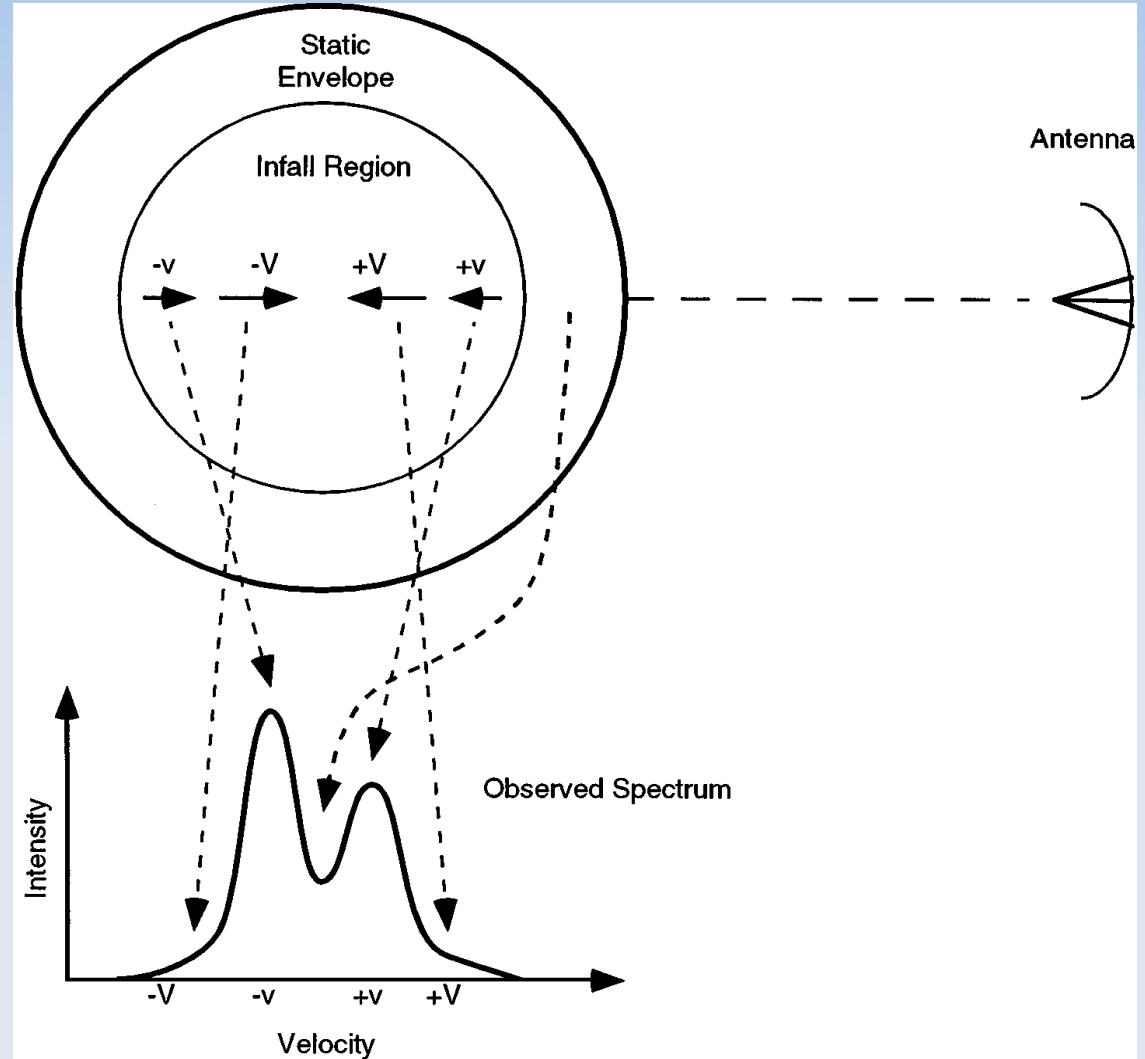
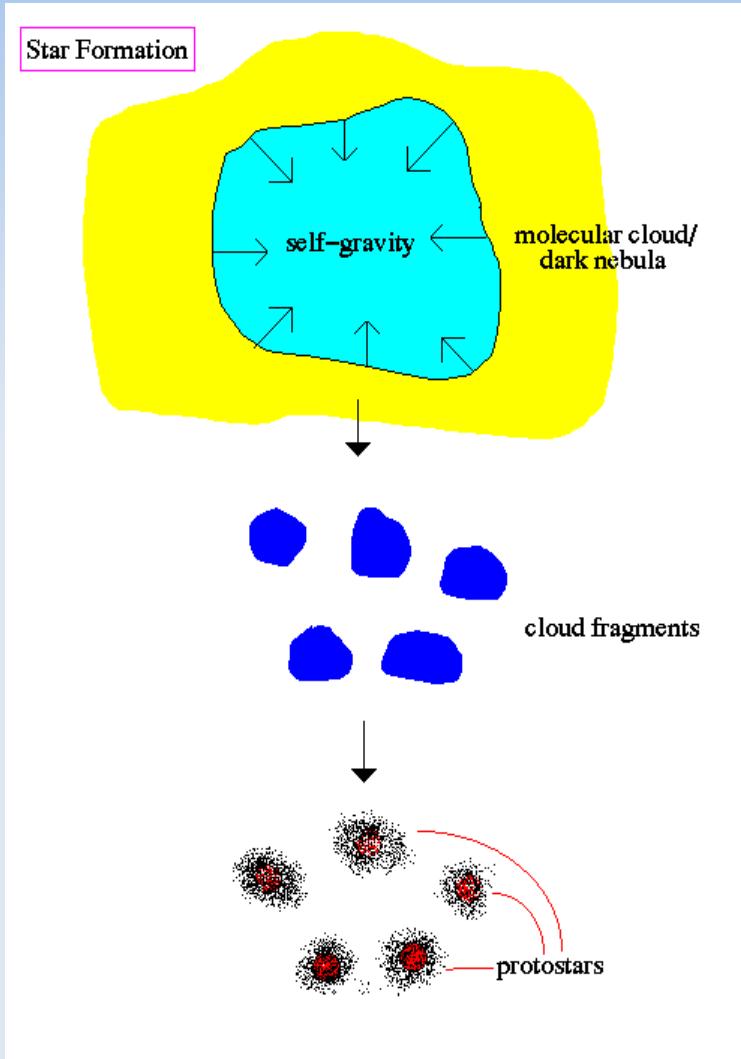
- Larger Doppler shifts can occur due to several effects:
 - Galactic rotation
 - Dynamical processes in molecular clouds and stars
 - Expansion of the universe (redshift can be so large that submm lines are shifted to cm wavelengths !)
- Each type of shift creates typical line profile shapes
- We often use radial Doppler velocity as x-axis



Dynamics: Rotation



Dynamics: Cloud Collapse

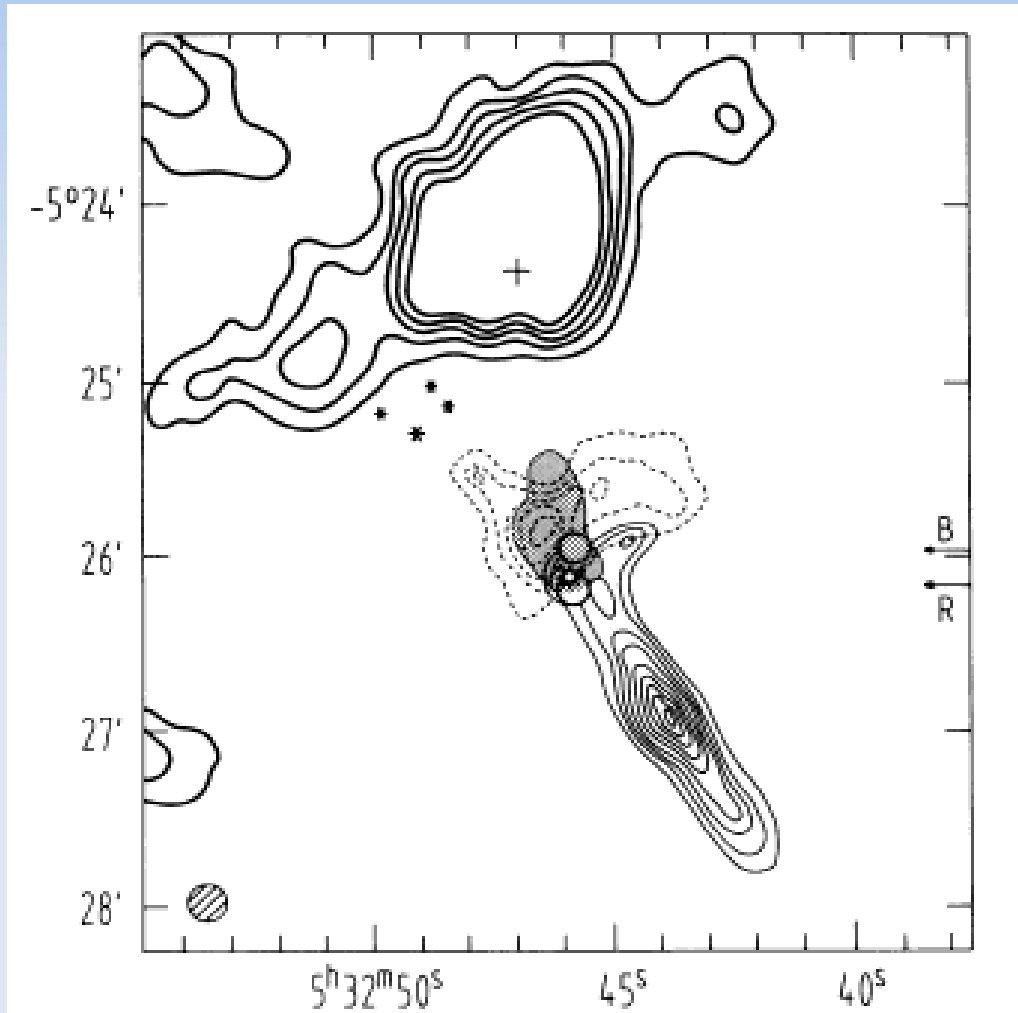


http://www.oglethorpe.edu/faculty/~m_rulison

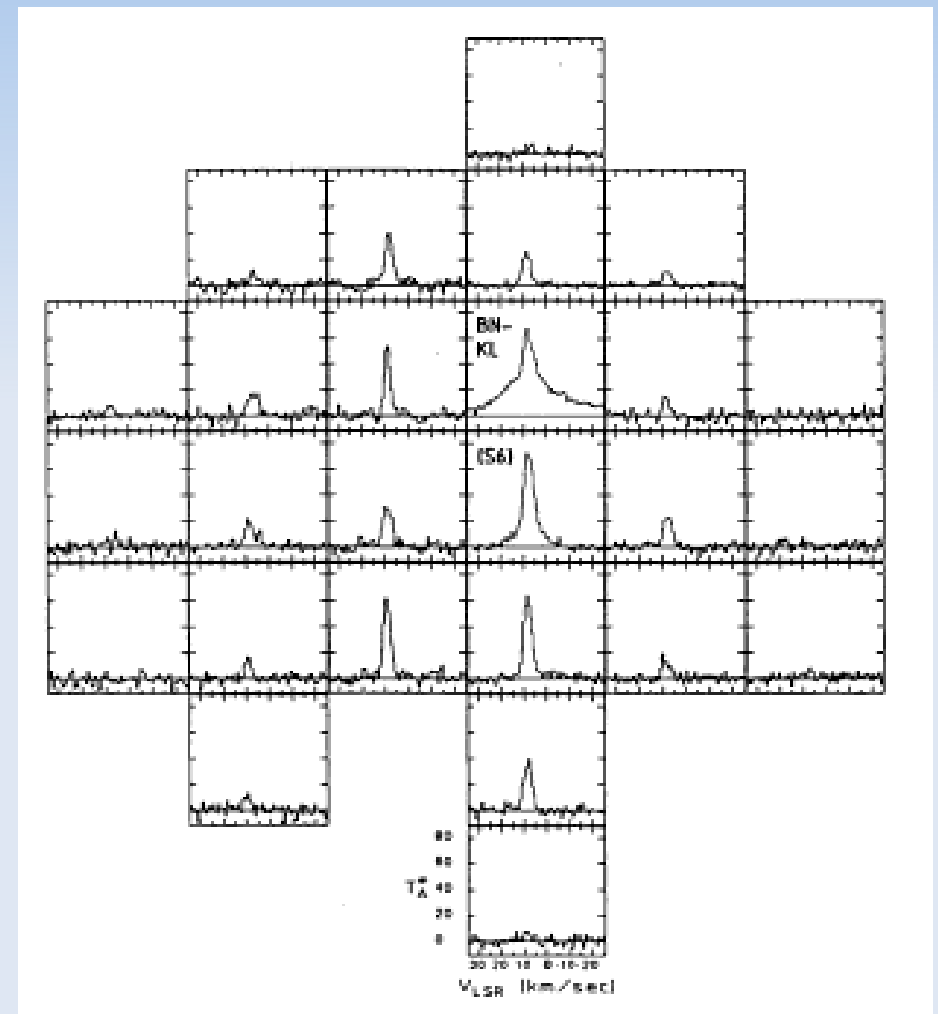
Neal J. Evans II, ARAA, 37, 311, 1999



Dynamics: Outflows



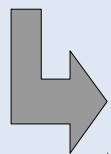
Schmid-Burgk et al. ApJ 362L, 25, 1990



Schmid-Burgk et al. LIACo, 29, 193, 1990

Spectral Instruments

- What does one need to observe radio astronomical spectral lines ?
 - Heterodyne frontend (→ special talk)
 - Usually a down-converter from observing frequencies to a “low” (0-4 GHz) intermediate frequency (IF) band
 - Spectrometers to analyze the signal



Will concentrate on spectrometers here



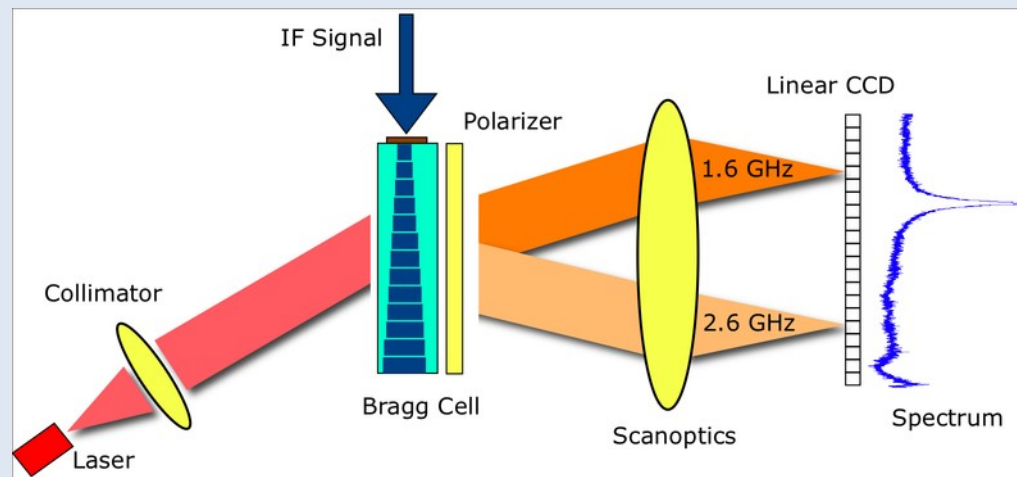
Spectrometers

- Spectrometers measure the frontend signal in many frequency bins across the available bandwidth
- There are several techniques:
 - Filter banks: Series of analog filters; complex electronics
 - Auto-correlators: Special purpose computers; correlation function of time series signals; low number of bits



Spectrometers (ctd.)

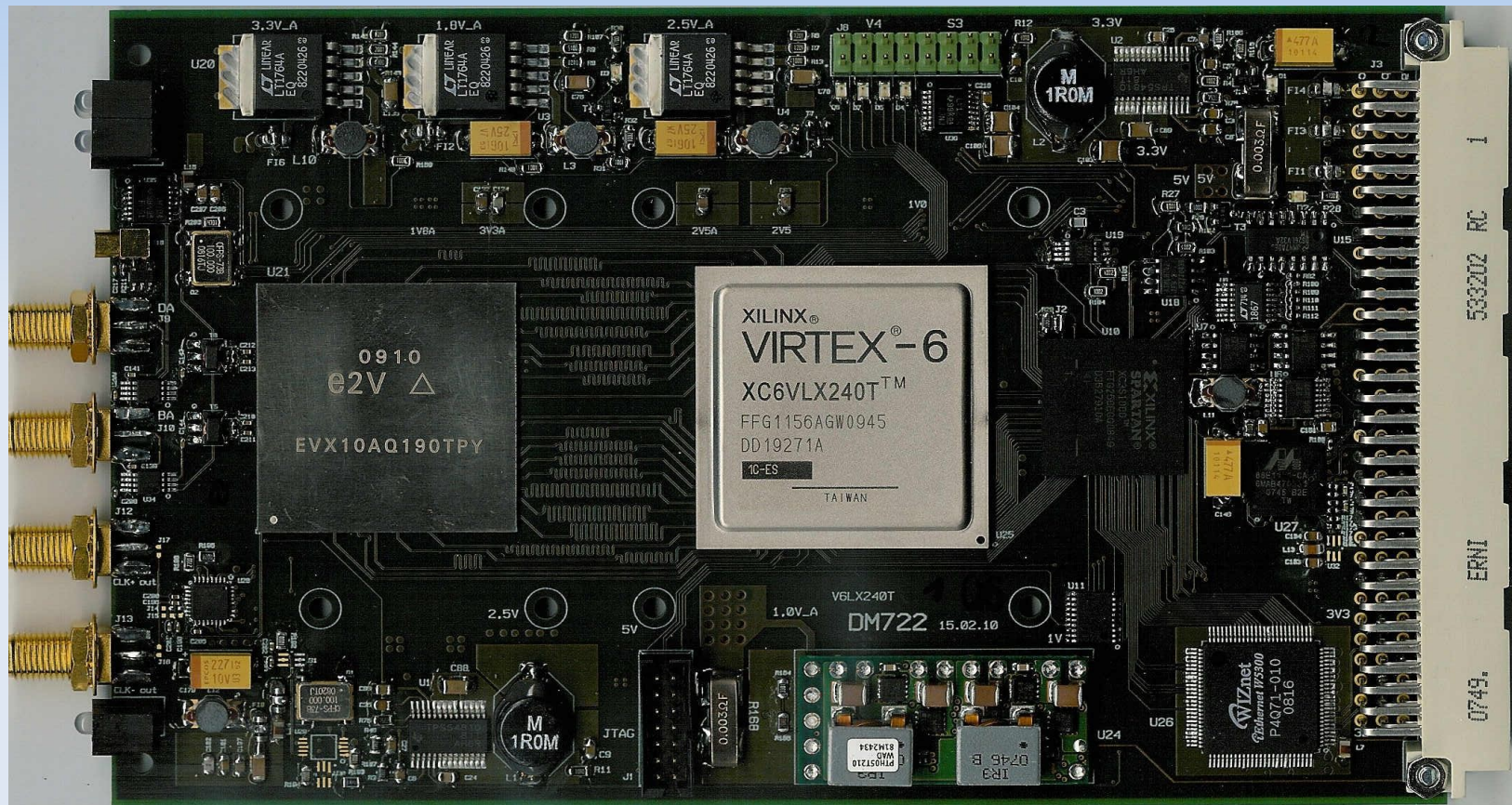
- More spectrometer types
 - Acousto-optical spectrometers (AOS): Diffraction of laser light at ultrasonic waves in a Bragg crystal; delicate optical setup



- Fast Fourier Transform Spectrometers: High speed ADCs and FPGAs → Development at the MPIfR

Fast Fourier Transform Spectrometer

XFFTS: 2.5 GHz bandwidth / 32768 channels (ENBW 88.5 kHz)



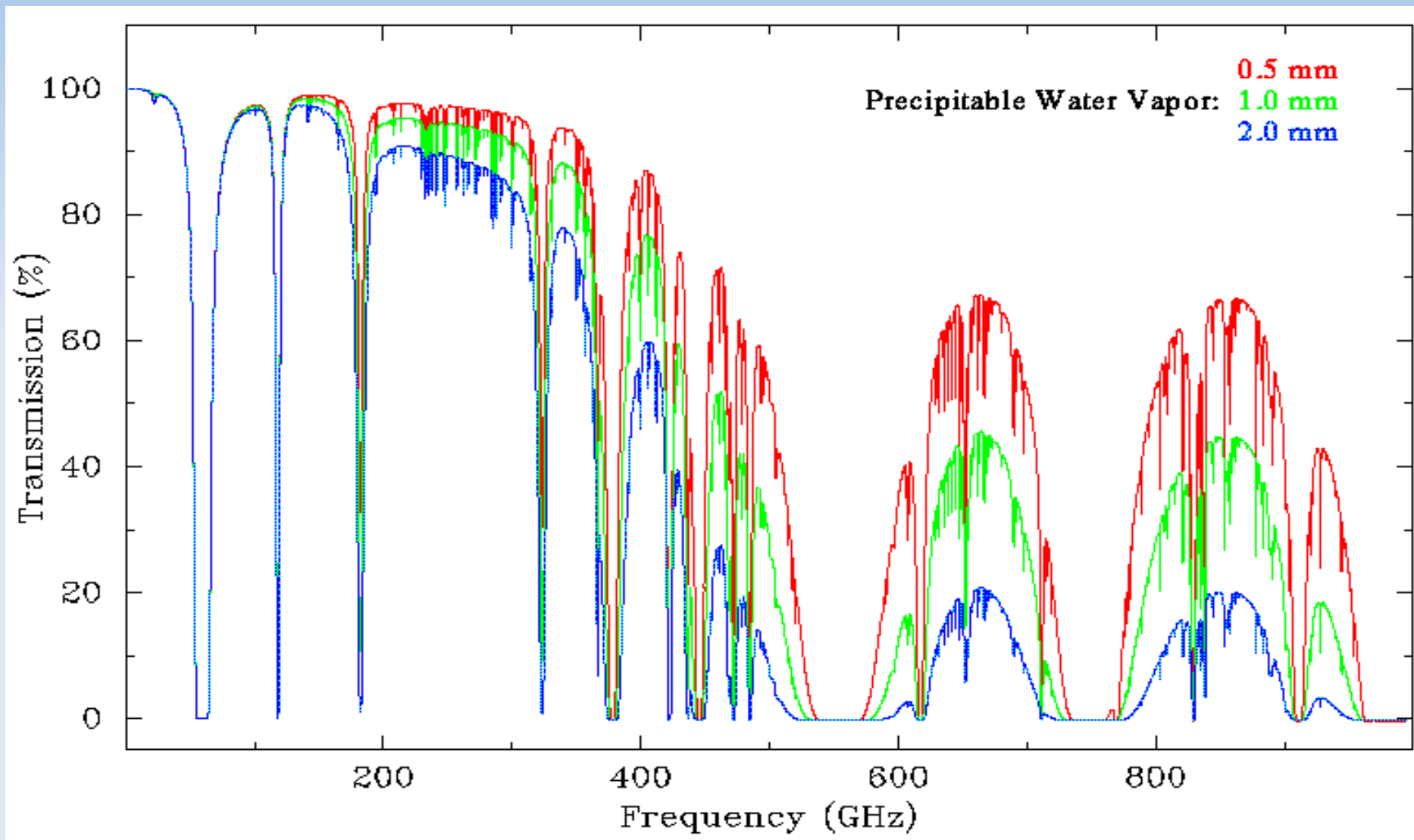
**E2V 5 GS/s 10-bit ADC, XILINX Virtex-6 LX240T
[40 nm, 1.0 volt core voltage, >240'000 logic cells, 768 DSP48 slices]**

Observations

- Source signal is partially absorbed by the earth atmosphere
- The atmosphere also radiates itself and thus contributes to the signal



Atmospheric Transmission



Observations

- Source signal is partially absorbed by the earth atmosphere
- The atmosphere also radiates itself and thus contributes to the signal
- The telescope beam picks up ground spillover
- Receiver etc. add a signal too
- Direct measurement therefore yields

$$C_{\text{on}} = C_{\text{source}} e^{-\tau A} + C_{\text{atm}} (1 - e^{-\tau A}) + C_{\text{spillover}} + C_{\text{rec}}$$



On-Off Technique

- To remove the atmospheric and instrumental emissions one observes the target and then a position on sky without astronomical emission

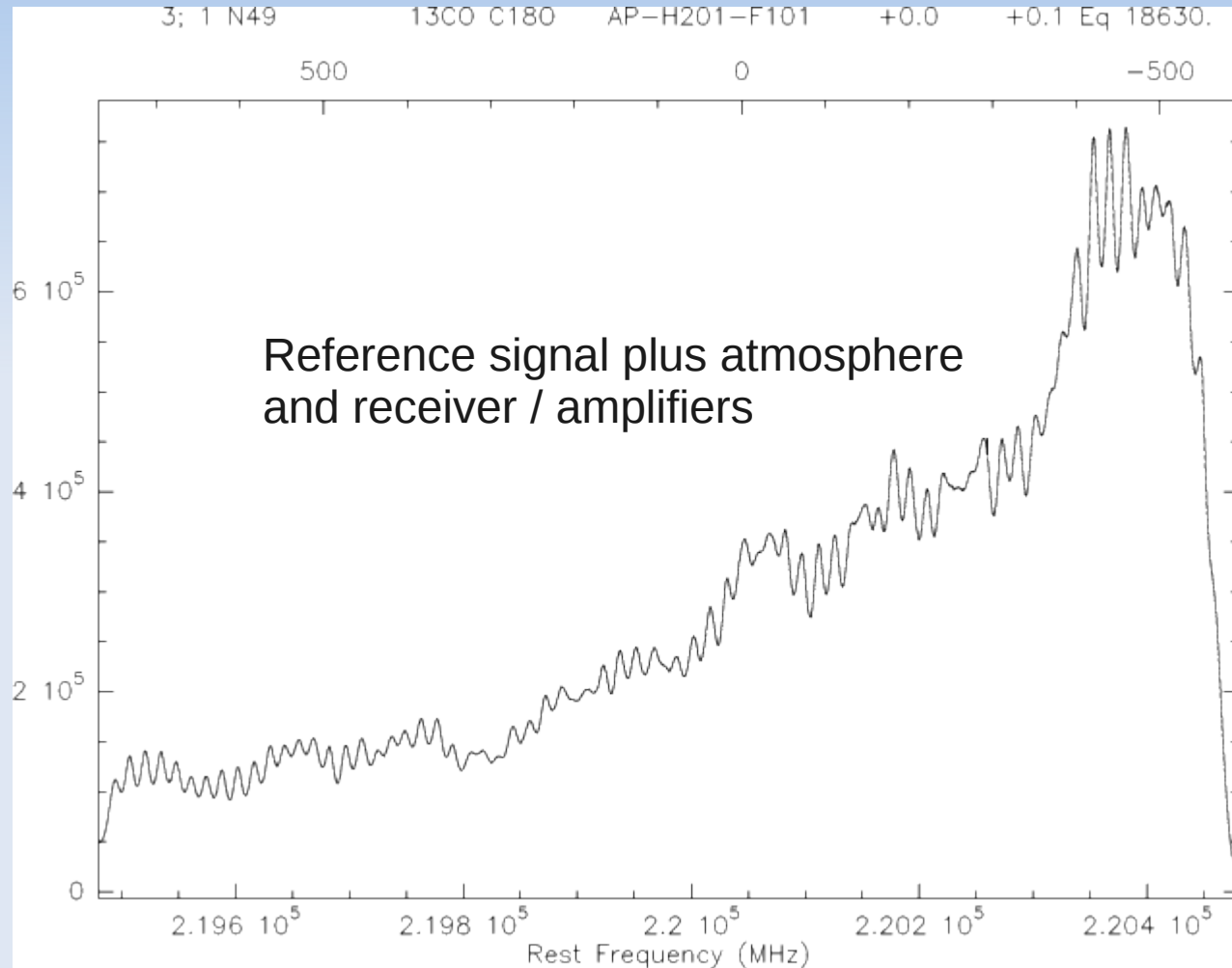
$$C_{\text{off}} = C_{\text{atm}} (1 - e^{-\tau A}) + C_{\text{spillover}} + C_{\text{rec}}$$

- The difference of the two measurements contains only the source signal (still weakened by atmospheric absorption):

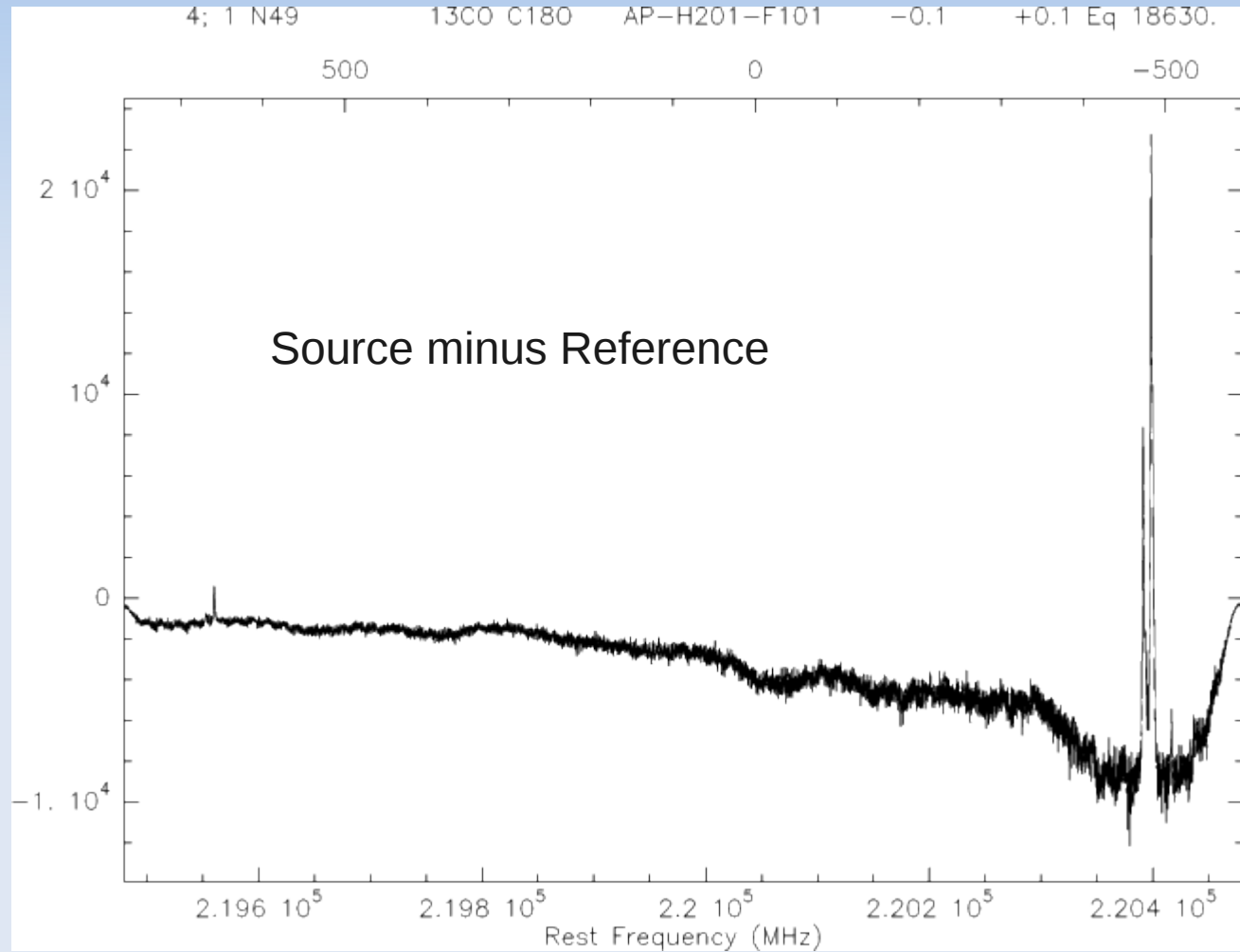
$$C_{\text{on}} - C_{\text{off}} = C_{\text{source}} e^{-\tau A}$$



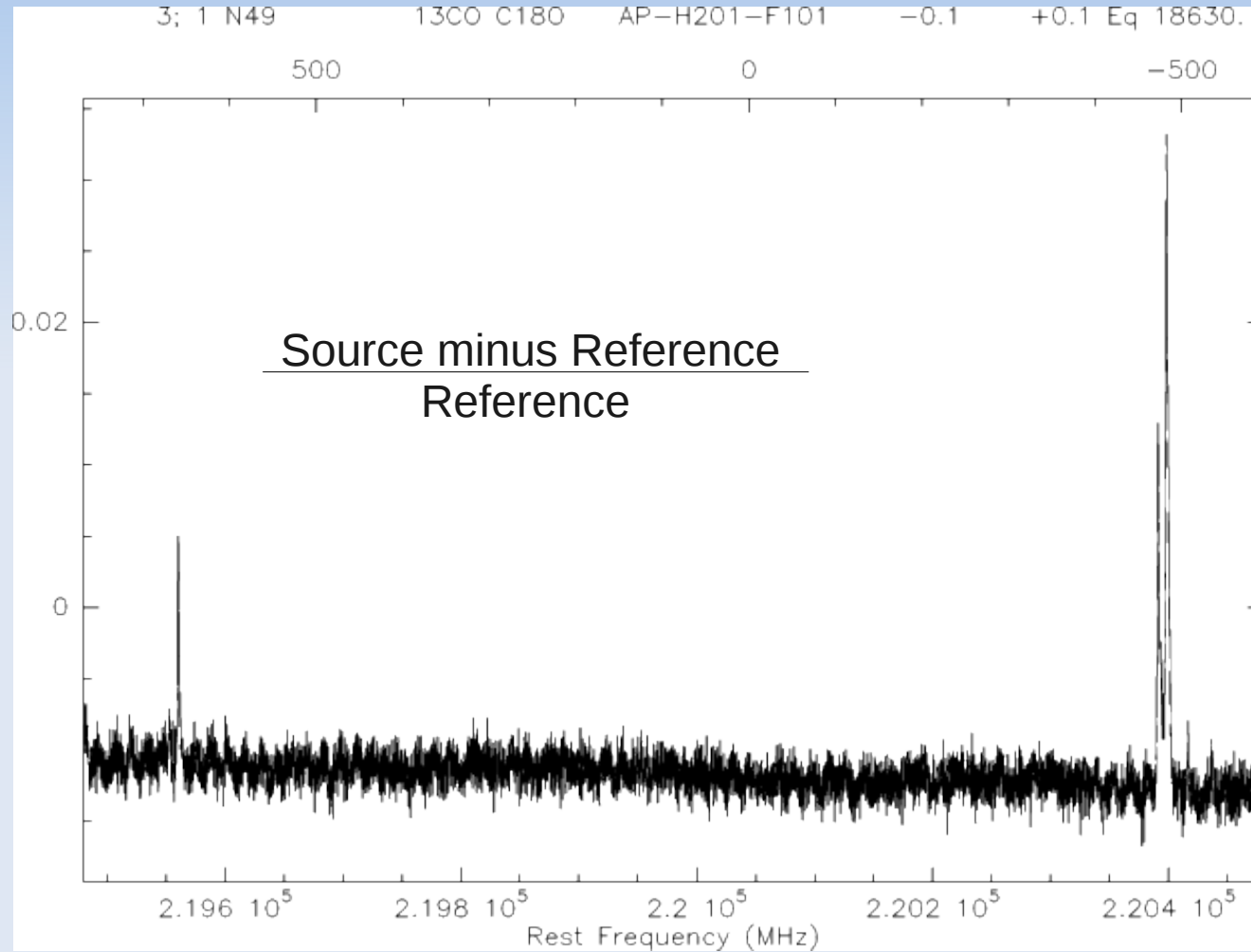
On-Off Technique



On-Off Technique



On-Off Technique



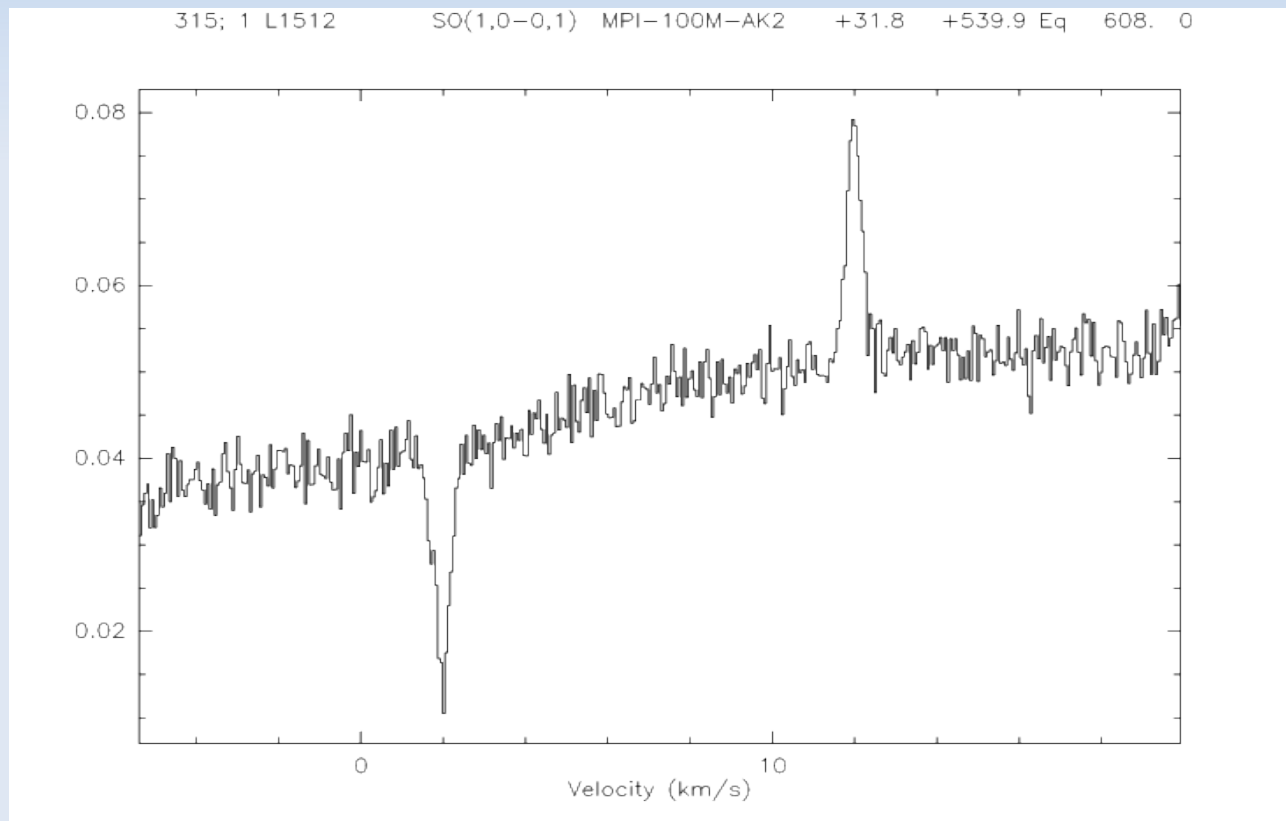
On-Off Alternatives

- On-Off measurements can be taken by moving the telescope between two positions.
- If the source is small, then one can use horn or wobbler switching which is faster. This helps if the atmospheric emission varies quickly.
- One can also measure the “off” at a slightly shifted frequency but pointing to the source. This doubles the actual “on” time and reduces telescope movements.



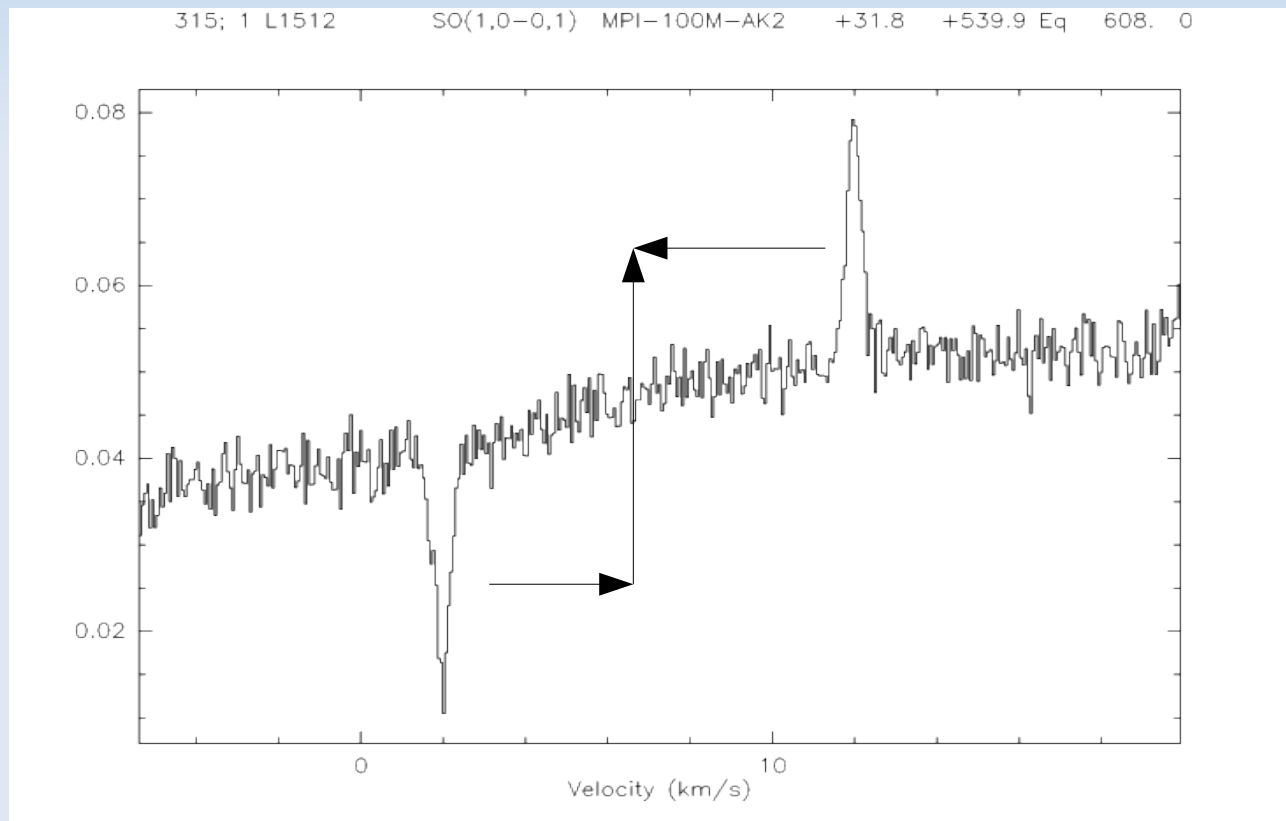
Frequency Switching

- Reference at a slightly different frequency but same position on the target



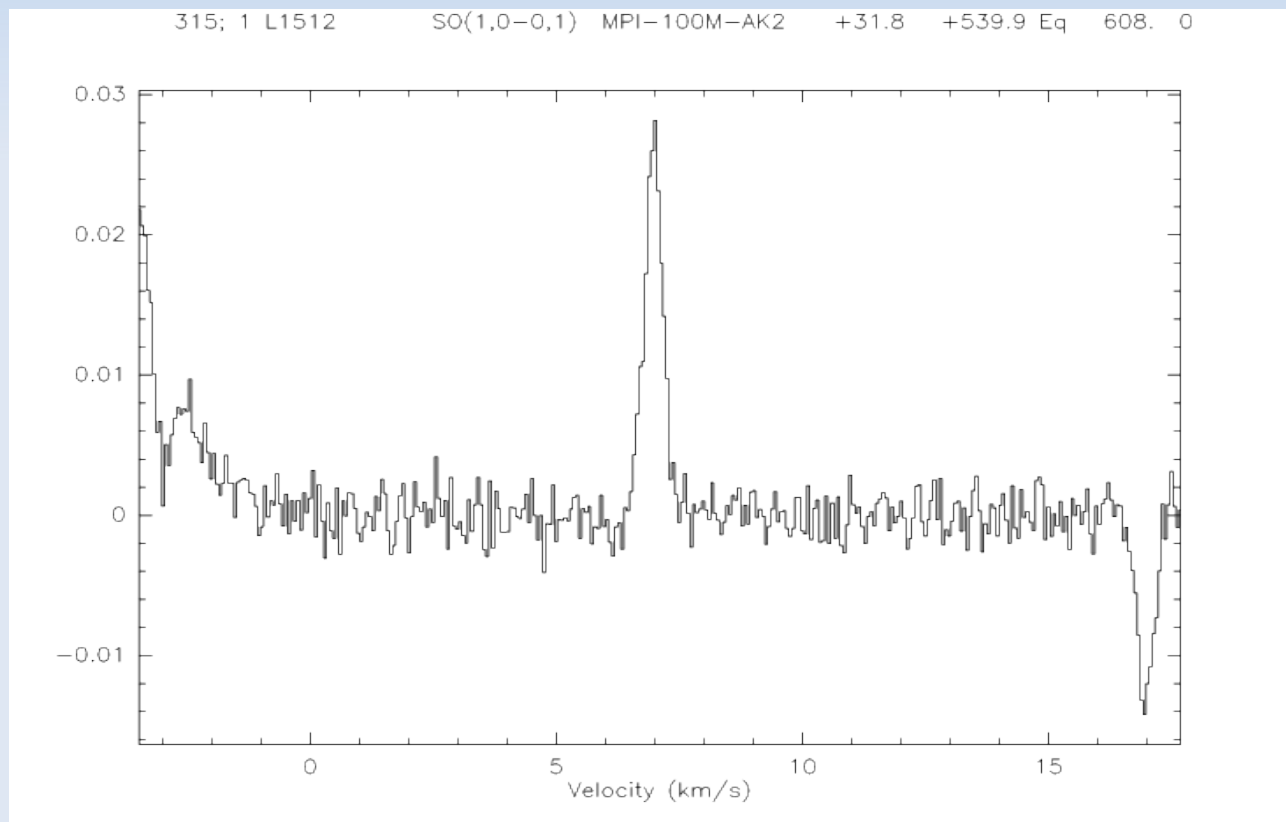
Frequency Switching

- Reference at a slightly different frequency but same position on the target



Frequency Switching

- Reference at a slightly different frequency but same position on the target

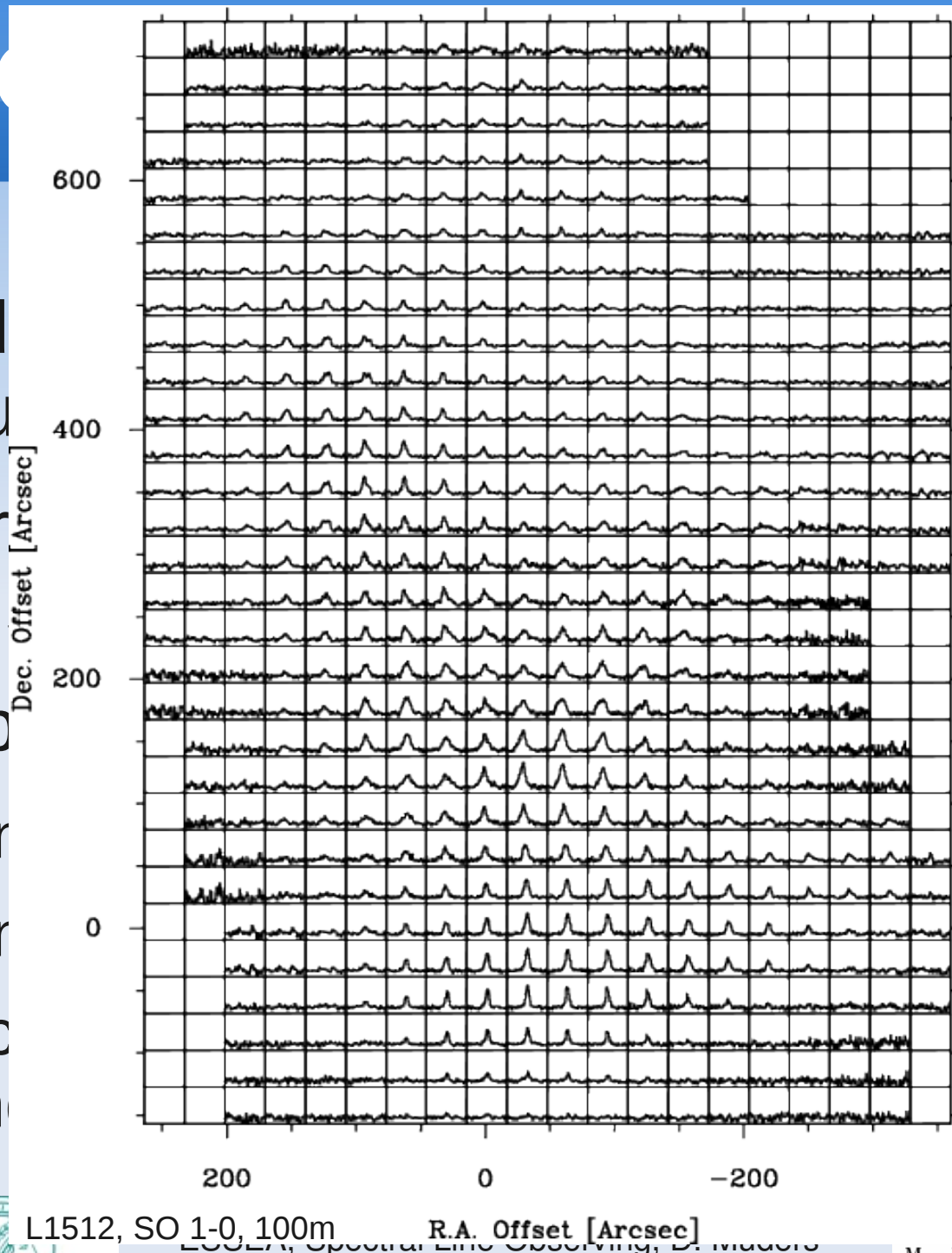


Observing Patterns

- Most radio receivers are still single pixel or small multi-beam systems
- To cover an extended source area one must observe several spatial offset positions
- Typical patterns are
 - (Rectangular) rasters with half beam spacing
 - (Rectangular) “On-The-Fly” rasters
 - Advanced figures like spirals, Lissajous figures, Rotating bow ties, etc.



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Calibration

- The measurements in arbitrary counts need to be calibrated to physical units
- For spectral lines one usually uses the “Antenna Temperature” scale
- The receiver system is calibrated against “hot” (usually ambient temperature) and “cold” (usually LN₂ @ 77 K) black bodies
- In cm wave receivers one uses a noise diode that was hot/cold calibrated in the lab



Calibration (ctd.)

- The absorption of astronomical signals needs to be corrected too:
 - Scaling according to measurements of secondary calibrator sources
 - Sky measurements and atmospheric models can be used to derive τ
 - More details in the calibration talk on Thursday



Data reduction

- Atmospheric and instrumental instabilities lead to spectral baseline artifacts



Data reduction

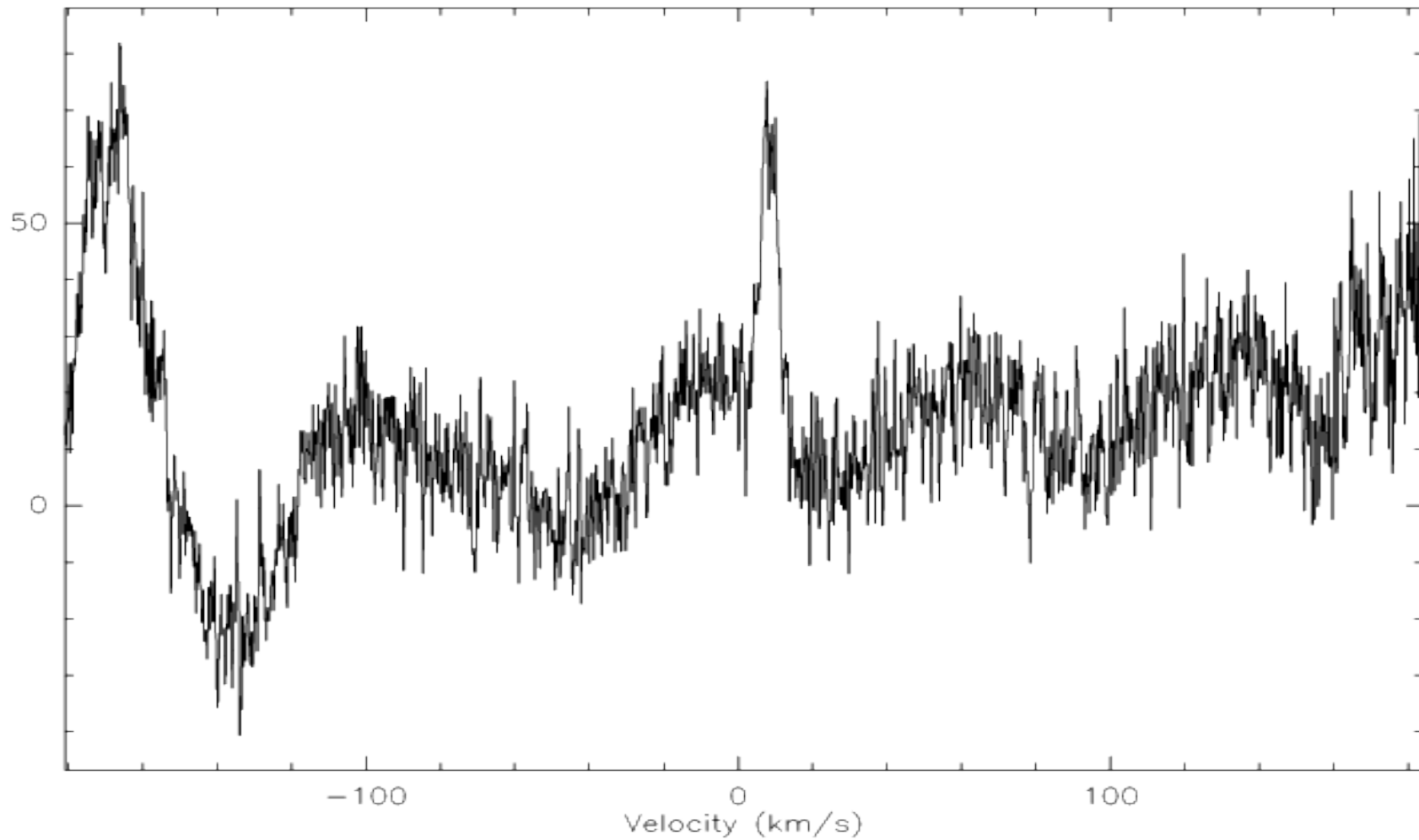
1109; 2 IRC2

CO(7-6)

SMT-10M-B40

-30.0

-28.4 Eq 3712. 0



Data reduction

- Atmospheric and instrumental instabilities lead to spectral baseline artifacts
- Techniques to process spectral line data include:
 - Spectral baseline fits using polynomials
 - FFT analysis to remove sinusoidal components due to standing waves
 - Flagging of very bad data
- Caveat: Must be very careful not to alter the line emission, esp. for broad lines



Data Products

- Primary data products are calibrated spectra
- For mapping projects the spatially distributed spectra are interpolated onto a regular grid to make 3D data cubes with two spatial and one spectral axis
- ALMA Pipeline Heuristics development (led by MPIfR) attempts to provide automatic data reduction (also applicable to Effelsberg data)

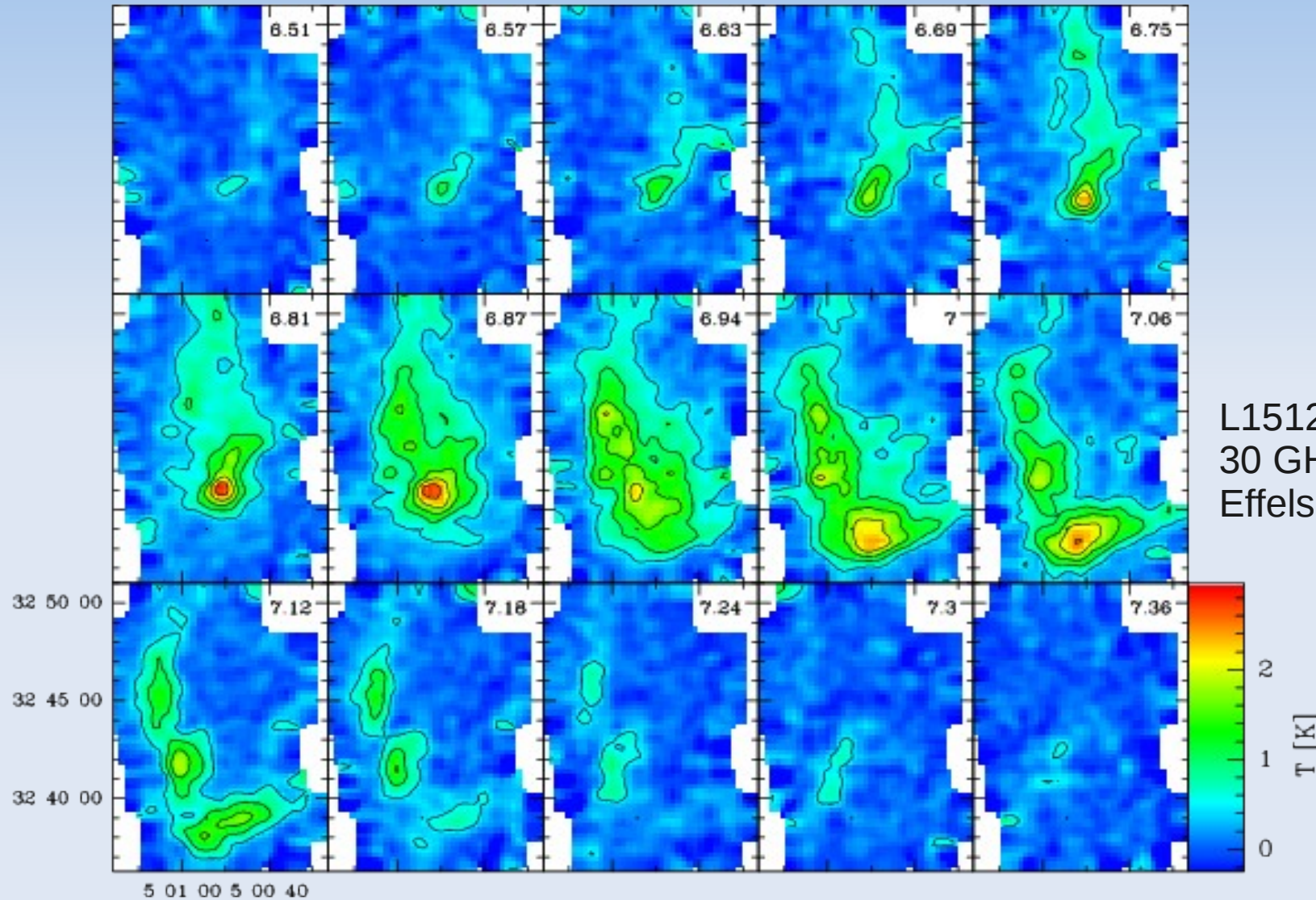


Spectral Data Visualization

- Usually display maps as false color images and contour plots
- Frequency axis allows for additional analysis, e.g. via so called channel maps or via position-velocity plots
- Since we have a data *cube*, one can apply 3D rendering techniques but one must be careful interpreting the graphs because of the frequency axis



Channel Maps



L1512, SO 1-0
30 GHz
Effelsberg 100m

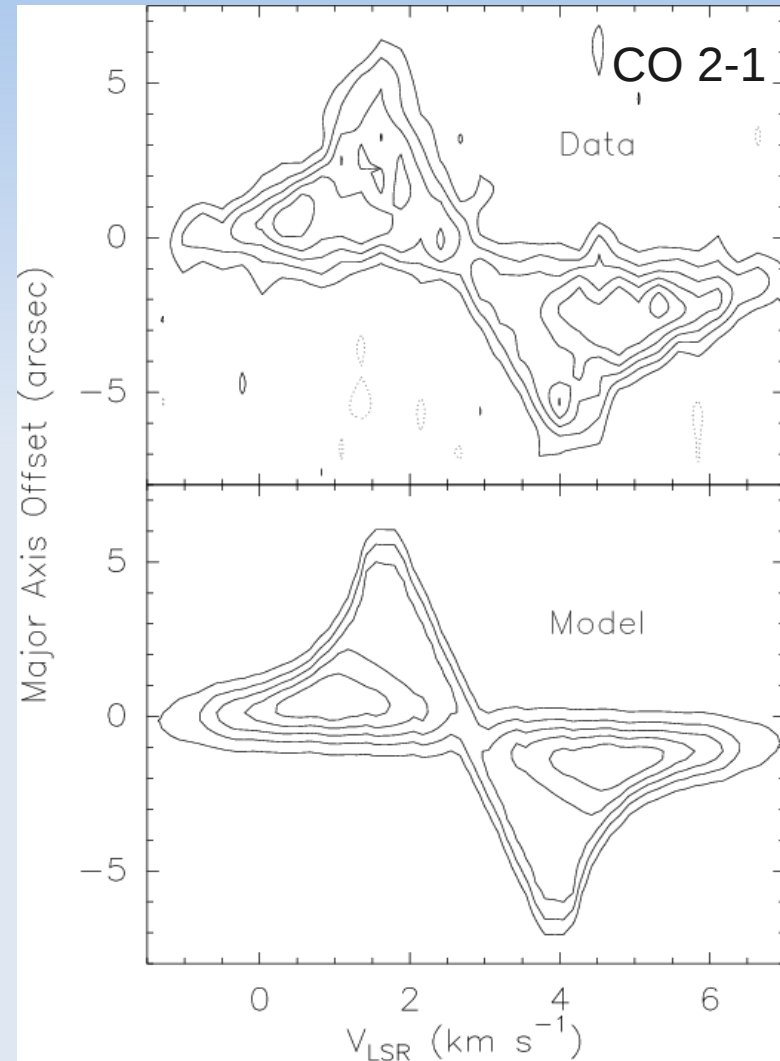


Position Velocity Plots

- Plot spatial axis against velocity to study cloud dynamics, e.g. Keplerian rotation



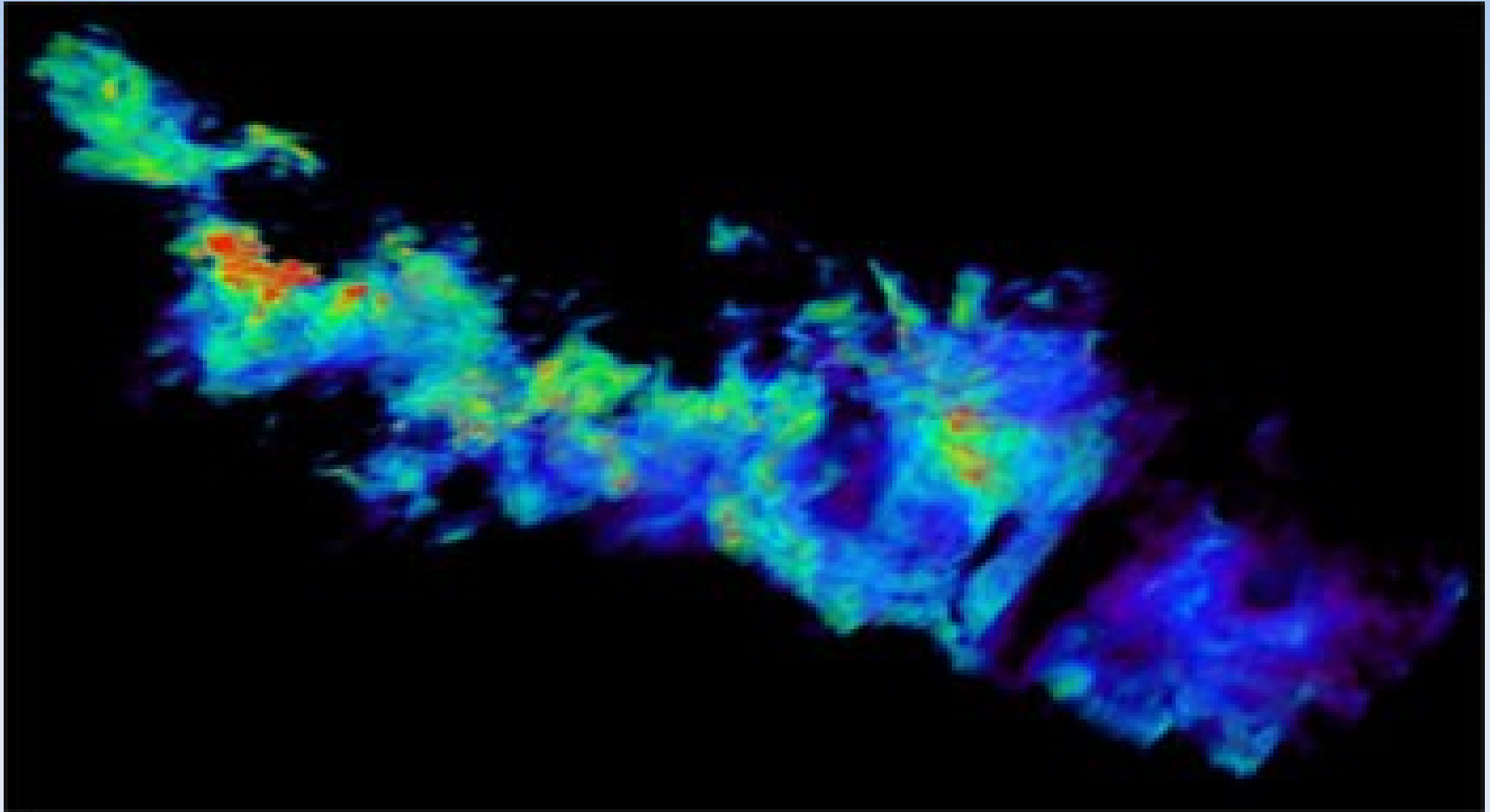
A. Gomez, CTIO, NOAO, HST, NASA



Bujarrabal et al. A&A 483, 839-845, 2008

Gomez's Hamburger (IRAS 18059-3211)

3D Rendering



2010-09-27



<http://am.iic.harvard.edu>
ESSEA, Spectral Line Observing, D. Muders






Max-Planck-Institut
für Radioastronomie

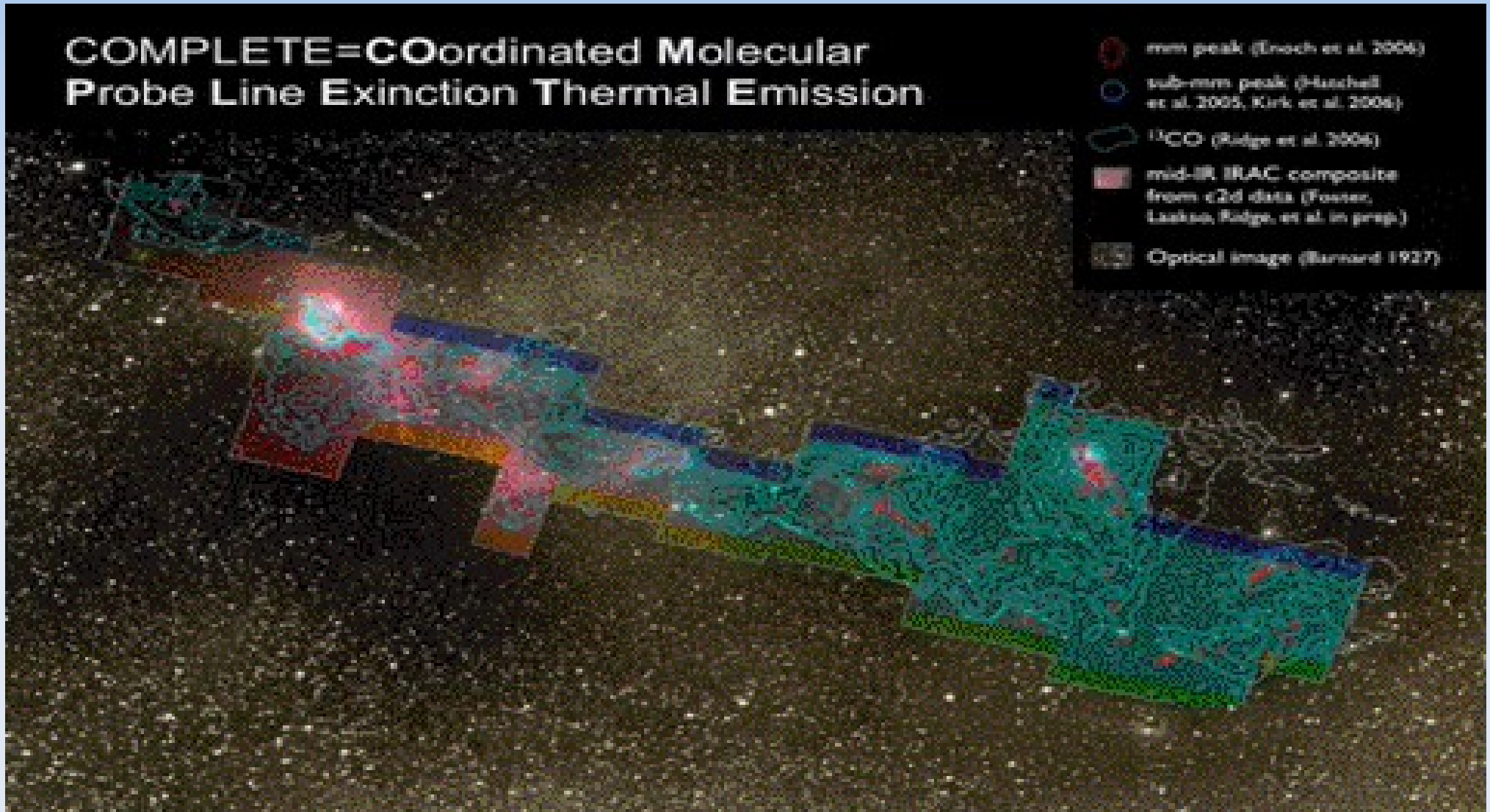


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3D Rendering

COMPLETE=COordinated Molecular
Probe Line Exinction Thermal Emission

-  mm peak (Enoch et al. 2006)
-  sub-mm peak (Hatchell et al. 2005, Kirk et al. 2006)
-  ^{13}CO (Ridge et al. 2006)
-  mid-IR IRAC composite from 2d data (Fosser, Laasaa, Ridge, et al. in prep.)
-  Optical image (Burnard 1927)



Deriving Physical Parameters

- Spectra and data cubes of several transitions are used in conjunction with models to derive physical parameter of the ISM:
 - Optically thin lines (involving isotopologues) to calculate column and volume densities
 - Line Ratios are modeled with chemical networks and radiative transfer programs
 - Spectral signatures of dynamical processes are fitted against the data
 - And many more ...

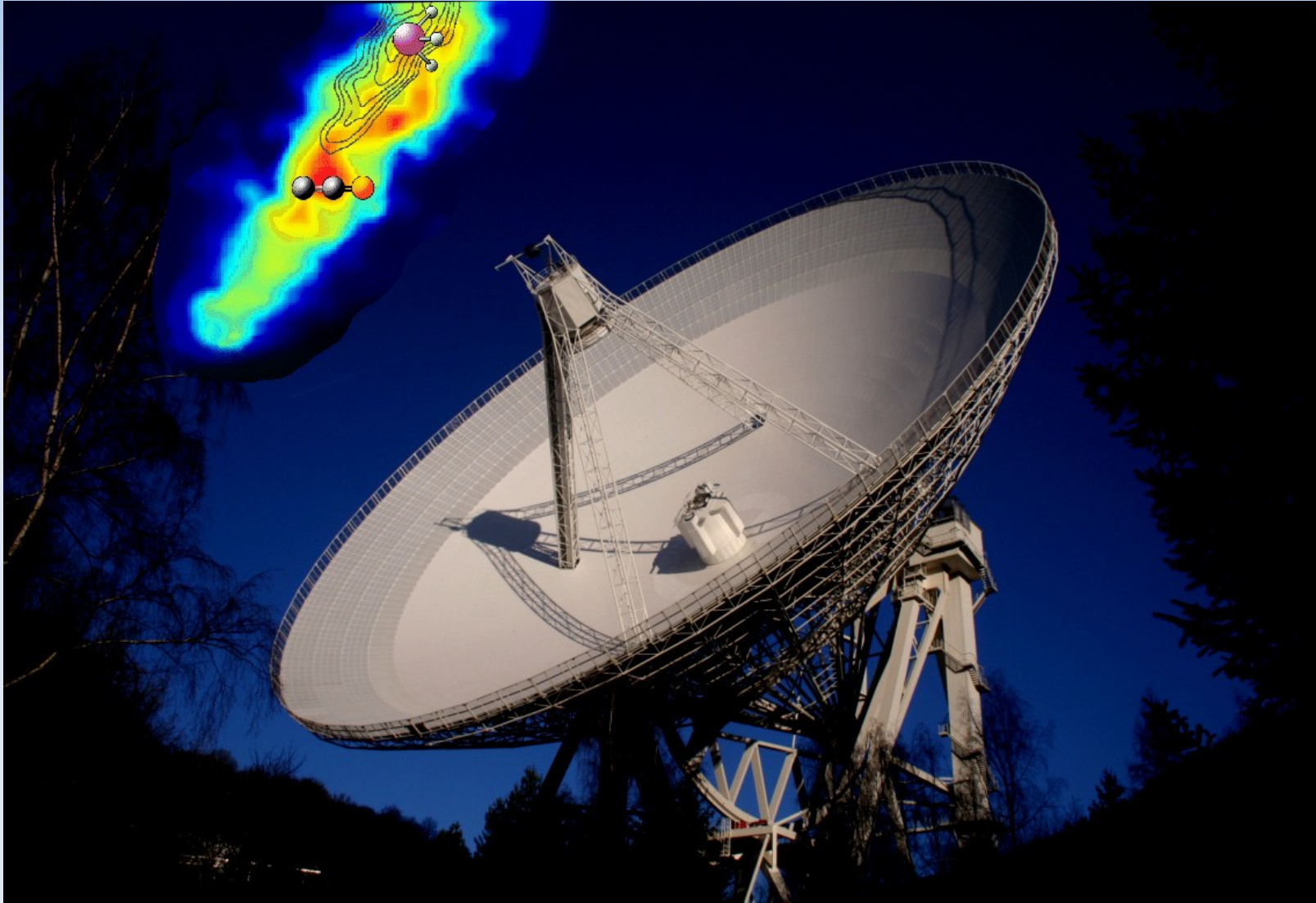


Summary

- Spectral lines provide a wealth of information about the interstellar medium
- Different atomic and molecular processes generate numerous spectral lines in the cm to submm wavelength range
- More than 160 molecules detected in space
- Physical, chemical and dynamical state of interstellar medium can be studied using spectral lines



Happy Observing !



2010-09-27



ESSEA, Spectral Line Observing, D. Muders

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für Radioastronomie



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Thermal Line Broadening

Since the thermal velocities are non-relativistic, the Doppler shift in the angular frequency is given by the simple form

$$\omega = \omega_0 \left(1 \pm \frac{v}{c} \right) \quad \omega_0 = \text{frequency for an atom at rest}$$

From the Boltzmann distribution, the number of atoms with velocity v in the direction of the observed light is given by

$$n(v)dv = N \sqrt{\frac{m_0}{2\pi kT}} e^{-m_0 v^2 / 2kT} dv \quad \begin{array}{l} N = \text{total number of atoms} \\ m_0 = \text{atomic mass} \end{array}$$

The distribution of radiation around the center frequency is then given by

$$I(\omega) = I_0 \exp \left[\frac{-m_0 c^2 (\omega_0 - \omega)^2}{2kT \omega_0^2} \right]$$

<http://hyperphysics.phy-astr.gsu.edu/hbase/atomic/broaden.html>



Antenna Temperature

Antenna temperature is a measure of signal strength in radio astronomy. It is defined as the temperature of a black-body enclosure which, if completely surrounding a radio telescope, would produce the same signal power as the source under observation. Antenna temperature is a property of the source, not of the antenna itself.

$$T_A = \frac{1}{4\pi} \int_0^{2\pi} \int_0^{\pi} R(\theta, \phi) T(\theta, \phi) \sin \theta d\theta d\phi$$

$R(\Theta, \Phi)$ is the antenna pattern.

